

GUIDED READING ACTIVITY



Pre-Reading Preparation

- Introducing the text and setting a purpose for reading
- Activating prior knowledge and making predictions
- Discussing vocabulary and key concepts

Reading the Text

- Reading aloud or silently, depending on the student's level
- Encouraging decoding strategies for unfamiliar words
- Monitoring comprehension through questioning

Post-Reading Discussion

- Summarizing key points
- Making connections to personal experiences or other texts
- Discussing themes, characters, and main ideas

Extension Activities

- Engaging in follow-up writing or drawing activities
- Using graphic organizers for comprehension
- Completing worksheets or interactive assignments

Guided Activity 2 answer

Richard Samuel Moog



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 $2 \text{ CO}_2(g) + \text{H}_2\text{O}(g) \rightarrow \text{C}_2\text{H}_2(g) + \frac{5}{2} \text{O}_2(g)$ $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(g) \rightarrow \text{H}_2(g) + \frac{1}{2} \text{O}_2(g)$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(g) + \frac{7}{2} \text{O}_2(g) \rightarrow \text{CO}_2(g) + 3 \text{H}_2\text{O}(g)$ $\Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(g) + 3 \text{O}_2(g) \rightarrow 2 \text{CO}_2(g) + 2 \text{H}_2\text{O}(l)$ $\Delta h = 1411 \text{ kJ}$
 $\text{C}_2\text{H}_6(g) + 7 \text{O}_2(g) \rightarrow 2 \text{CO}_2(g) + 3 \text{H}_2\text{O}(l)$ $\Delta h = 1560 \text{ kJ}$
 $\text{H}_2(g) + \frac{1}{2} \text{O}_2(g) \rightarrow \text{H}_2\text{O}(l)$ $\Delta h = 285 \text{ kJ}$
 mole

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given the following information

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web thus for reaction 9 2 in which 1 mole of H_2SO_4 and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta H = 55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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