

Subatomic  
Physics

## HOLT PHYSICS

**Concept Review***The Nucleus*

1. A certain atom has eight protons, eight electrons, and eight neutrons.
  - a. How many nucleons does this atom have?  
\_\_\_\_\_
  - b. What is the atomic number of this atom?  
\_\_\_\_\_
  - c. What is the mass number of this atom?  
\_\_\_\_\_
  - d. If the nucleus of this atom has a mass of 16.124 552 u, calculate the binding energy of the nucleus.  
\_\_\_\_\_
  - e. What is the significance of the binding energy?  
\_\_\_\_\_  
\_\_\_\_\_
  - f. Would an atom with eight protons, eight electrons, and nine neutrons be a different element? Explain.  
\_\_\_\_\_  
\_\_\_\_\_
2. Two protons in a nucleus experience a very large repulsion force.
  - a. What prevents these two protons from accelerating away from each other?  
\_\_\_\_\_
  - b. As a nucleus gets larger, what happens to the ratio of protons to neutrons?  
\_\_\_\_\_  
\_\_\_\_\_

# Holt Study Guide Subatomic Physics Concept Answers

**S Marginson**



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