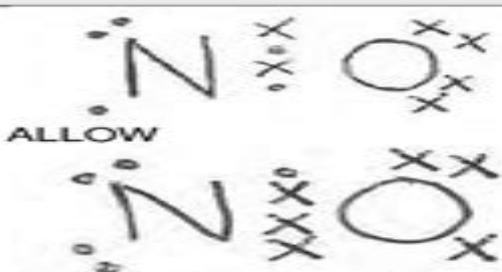


## Section B

Question Number	Acceptable Answers	Reject	Mark
<b>19(a)</b>	(in $\text{NH}_3 =$ ) -3/3-/-III (1) (in $\text{NO} =$ ) +2/2+ /+ II (1)	Just '2'	2
Question Number	Acceptable Answers	Reject	Mark
<b>19(b)(i)</b>	It has an unpaired electron  ALLOW non-paired  Ignore references to reactivity/ stability/ orbital/charge/location of unpaired electron	Just 'single electron' 'lone electron'  Electrons Free electron	1
Question Number	Acceptable Answers	Reject	Mark
<b>19(b)(ii)</b>	 <p>ALLOW</p> <p>Double bond as shown in either of above diagrams (1)</p> <p>2 lone pairs on one atom and 1 lone pair + 1 unpaired electron on the other atom (1)</p> <p>Second mark dependent on the first</p>		2

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**Ben Sparks**



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