

# CHAPTER 14 STUDY GUIDE

## Mixtures and Solutions

### Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

- false  1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed.
- true  2. The most abundant substance in a colloid is the dispersion medium.
- false  3. A colloid can be separated by filtration.
- true  4. A solid emulsion consists of a liquid dispersed in a solid.
- true  5. Whipped cream is an example of a foam.
- false  6. In an aerosol, the dispersing medium is a liquid.
- true  7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
- true  8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
- false  9. Stirring an electrolyte into a colloid stabilizes the colloid.
- true  10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

| Characteristics of Particles             | Colloid | Solution | Suspension |
|--|---------|----------|------------|
| 11. Less than 1 nm in diameter           |         | ✓        | ✓          |
| 12. Between 1 nm and 1000 nm in diameter | ✓       | ✓        | ✓          |
| 13. More than 1000 nm in diameter        | ✓       | ✓        |            |
| 14. Settle out if undisturbed            |         |          |            |
| 15. Pass through standard filter paper   | ✓       |          |            |
| 16. Lower vapor pressure                 |         |          | ✓          |
| 17. Scatter light                        | ✓       |          |            |

# Mixtures And Solutions Study Guide Answer Key

**Zhe Xu**



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## **Table of Contents Mixtures And Solutions Study Guide Answer Key**

1. Understanding the eBook Mixtures And Solutions Study Guide Answer Key
  - The Rise of Digital Reading Mixtures And Solutions Study Guide Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Mixtures And Solutions Study Guide Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mixtures And Solutions Study Guide Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mixtures And Solutions Study Guide Answer Key
  - Personalized Recommendations
  - Mixtures And Solutions Study Guide Answer Key User Reviews and Ratings
  - Mixtures And Solutions Study Guide Answer Key and Bestseller Lists
5. Accessing Mixtures And Solutions Study Guide Answer Key Free and Paid eBooks
  - Mixtures And Solutions Study Guide Answer Key Public Domain eBooks
  - Mixtures And Solutions Study Guide Answer Key eBook Subscription Services
  - Mixtures And Solutions Study Guide Answer Key Budget-Friendly Options
6. Navigating Mixtures And Solutions Study Guide Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Mixtures And Solutions Study Guide Answer Key Compatibility with Devices
  - Mixtures And Solutions Study Guide Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mixtures And Solutions Study Guide Answer Key
  - Highlighting and Note-Taking Mixtures And Solutions Study Guide Answer Key
  - Interactive Elements Mixtures And Solutions Study Guide Answer Key
8. Staying Engaged with Mixtures And Solutions Study Guide Answer Key

- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mixtures And Solutions Study Guide Answer Key
9. Balancing eBooks and Physical Books Mixtures And Solutions Study Guide Answer Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mixtures And Solutions Study Guide Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Mixtures And Solutions Study Guide Answer Key
- Setting Reading Goals Mixtures And Solutions Study Guide Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mixtures And Solutions Study Guide Answer Key
- Fact-Checking eBook Content of Mixtures And Solutions Study Guide Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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