

Section 4.5

Types of Aqueous Solutions and Solubility

Solubility Rules for Ionic Compounds in Water

TABLE 4.1 Solubility Rules for Ionic Compounds in Water	
Compounds Containing the Following Ions Are Generally Soluble	Exceptions
Li^+ , Na^+ , K^+ , and NH_4^+	None
NO_3^- and $\text{C}_2\text{H}_3\text{O}_2^-$	None
Cl^- , Br^- , and I^-	When these ions pair with Ag^+ , Hg_2^{2+} , or Pb^{2+} , the resulting compounds are insoluble.
SO_4^{2-}	When SO_4^{2-} pairs with Sr^{2+} , Ba^{2+} , Pb^{2+} , Ag^+ , or Ca^{2+} , the resulting compound is insoluble.
Compounds Containing the Following Ions Are Generally Insoluble	Exceptions
OH^- and S^{2-}	When these ions pair with Li^+ , Na^+ , K^+ , or NH_4^+ , the resulting compounds are soluble.
	When S^{2-} pairs with Ca^{2+} , Sr^{2+} , or Ba^{2+} , the resulting compound is soluble.
	When OH^- pairs with Ca^{2+} , Sr^{2+} , or Ba^{2+} , the resulting compound is slightly soluble.
CO_3^{2-} and PO_4^{3-}	When these ions pair with Li^+ , Na^+ , K^+ , or NH_4^+ , the resulting compounds are soluble.

Model Rules Of Solubility In Aqueous Solutions Pogil

John Henry Saylor



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Handbook of Aqueous Solubility Data Samuel H. Yalkowsky, Yan He, Parijat Jain, 2016-04-19 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until the first publication of the Handbook of Aqueous Solubility Data this information had been scattered throughout numerous sources Now newly revised the second edition of **Handbook of Aqueous Solubility Data** Samuel H. Yalkowsky, 2003-03-26 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until now this information has been scattered throughout the literature Containing over 16 000 solubility data points for more than 4 000 organic compounds **Handbook of Aqueous Aqueous Solution and the Phase Diagram** Frederick Field Purdon, Victor Wallace Slater, 1946 *The Experimental Determination of Solubilities* G. T. Hefter, R. P. T. Tomkins, 2003-08-01 In diesem Band werden vielfältige experimentelle Verfahren zur Bestimmung von Löslichkeiten beschrieben Commission on Solubility Data der IUPAC beschlo die Qualität der Löslichkeitsdaten international anzugleichen und zu verbessern indem verlässliche Verfahren zur Ermittlung dieser Daten zusammengefasst veröffentlicht werden alle Datentypen wurden hier berücksichtigt ausführliche Angaben zur Zuverlässigkeit der einzelnen Verfahren und zur Auswahl der jeweils geeigneten Methode *A Study of the Solubility of Certain Organic Substances in Aqueous Solutions* John Henry Saylor, 1930 *Solubility Behavior of Pharmaceuticals in Aqueous Solutions* Mohit Chandra Gupta, 1952 *Solubilities of Inorganic and Organic Substances* Atherton Seidell, 1928 **Solubilities of Inorganic and Organic Compounds** Atherton Seidell, 1919 **Developments and Applications in Solubility** Trevor M Letcher, 2007-10-31 Solubility is fundamental to most areas of chemistry and is one of the most basic of thermodynamic properties It underlies most industrial processes Bringing together the latest developments and ideas *Developments and Applications in Solubility* covers many varied and disparate topics The book is a collection of work from leading experts in their fields and covers the theory of solubility modelling and simulation industrial applications and new data and recent developments relating to solubility Of particular interest are sections on experimental calculated and predicted solubilities solubility phenomena in green quaternary mixtures involving ionic liquids molecular simulation approaches to solubility solubility impurities in cryogenic liquids and carbon dioxide in chemical processes The book is a definitive and comprehensive reference to what is new in solubility and is ideal for researcher scientists industrialists and academics **Solubilities of Inorganic and Organic Substances**, 1907 **Solubility in Inorganic Two-component Systems** Miroslav Broul, 1981 **Solubility and Solubilization in Aqueous Media** Samuel Hyman Yalkowsky, 1999 This book aims to provide the reader with a working knowledge of the various means of controlling the solubility or dissolution rate of a drug or other solute in an aqueous medium The book begins with the factors which govern solubility in general and then looks at aqueous solubility in particular including the properties of liquid mixtures and the

thermodynamics of solutions formed from mixing two components The bulk of the book is then devoted to techniques for altering solubility and dissolution rate of organic compounds in aqueous media It discusses in detail the most commonly used solubility enhancers buffers cosolvents surfactants and complexants Each chapter is self contained and emphasizes the details for applying the techniques

Solubility in Inorganic Two-component Systems Miroslav Broul, Jaroslav Nyvlt, Otakar Söhnel, 1981

Solubilities of inorganic and organic compounds Atherton Seidell, William F. Linke, 1952

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