

Unit 6 – Worksheet 3

Ionic Compounds

Properties

Basic structural unit

1. Give the name of the following simple binary ionic compounds.

- Na_2O
- K_2S
- MgCl_2
- CaBr_2
- BaI_2
- Al_2S_3
- CaBr
- AgF

2. Give the name of the following simple binary ionic compounds.

- Na_3N
- K_2O
- AgBr
- MgI_2
- SrO

3. Write the formula for the following binary ionic compounds.

- lithium bromide
- sodium iodide
- silver sulfide
- cesium oxide
- beryllium iodide
- barium hydride
- aluminum fluoride
- potassium oxide

4. Write the formula for these ionic substances.

- silver oxide
- aluminum sulfide
- sodium nitride
- barium chloride
- strontium hydride

Modeling Chemistry Unit Ws 2

University of California, San Diego

A red circular graphic with a gradient, appearing as a partial circle on the right side of the slide, overlapping the light blue bar.

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