

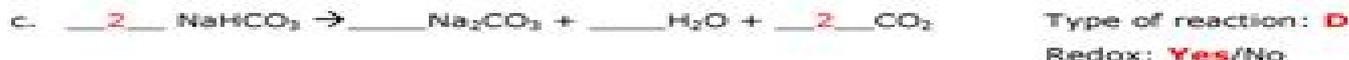
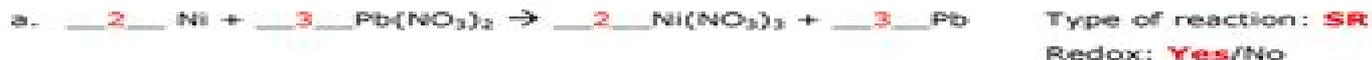
## Chemistry Unit 3 Review

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1. Define the Law of Conservation of Mass

**Mass is not created or destroyed.**

2. Using your knowledge of the types of chemical reaction, balance the following equations and indicate which of the six types of reaction they are:



3. Define oxidation-reduction (redox) reaction:

**The charges change from the reactant to product side of the reaction. Oxidation loses electrons- it gets more positive. Reduction gains electrons- it gets more negative.**

4. Given the following word equations, write the complete balanced chemical equation and identify the type of reaction.

- a) Copper (II) chloride reacts with iron to produce iron (III) chloride and copper metal.



- b) Hydrogen gas and bromine liquid react to yield hydrogen bromide.



- c) In a Bunsen burner, carbon tetrahydride (methane gas) reacts with oxygen to produce carbon dioxide and water vapor.



# Modeling Chemistry Unit 3 Review Answers

**J Elliott**



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