

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	56	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

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Table of Contents Mcdonald Publishing Co Basic Chemistry Answer Key

1. Understanding the eBook Mcdonald Publishing Co Basic Chemistry Answer Key
 - The Rise of Digital Reading Mcdonald Publishing Co Basic Chemistry Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcdonald Publishing Co Basic Chemistry Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcdonald Publishing Co Basic Chemistry Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcdonald Publishing Co Basic Chemistry Answer Key

- Personalized Recommendations
 - Mcdonald Publishing Co Basic Chemistry Answer Key User Reviews and Ratings
 - Mcdonald Publishing Co Basic Chemistry Answer Key and Bestseller Lists
5. Accessing Mcdonald Publishing Co Basic Chemistry Answer Key Free and Paid eBooks
 - Mcdonald Publishing Co Basic Chemistry Answer Key Public Domain eBooks
 - Mcdonald Publishing Co Basic Chemistry Answer Key eBook Subscription Services
 - Mcdonald Publishing Co Basic Chemistry Answer Key Budget-Friendly Options
 6. Navigating Mcdonald Publishing Co Basic Chemistry Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Mcdonald Publishing Co Basic Chemistry Answer Key Compatibility with Devices
 - Mcdonald Publishing Co Basic Chemistry Answer Key Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcdonald Publishing Co Basic Chemistry Answer Key
 - Highlighting and Note-Taking Mcdonald Publishing Co Basic Chemistry Answer Key
 - Interactive Elements Mcdonald Publishing Co Basic Chemistry Answer Key
 8. Staying Engaged with Mcdonald Publishing Co Basic Chemistry Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcdonald Publishing Co Basic Chemistry Answer Key
 9. Balancing eBooks and Physical Books Mcdonald Publishing Co Basic Chemistry Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcdonald Publishing Co Basic Chemistry Answer Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Mcdonald Publishing Co Basic Chemistry Answer Key
 - Setting Reading Goals Mcdonald Publishing Co Basic Chemistry Answer Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Mcdonald Publishing Co Basic Chemistry Answer Key

- Fact-Checking eBook Content of Mcdonald Publishing Co Basic Chemistry Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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