

Unit 6 – Worksheet 3

Ionic Compounds

Properties

Basic structural unit

1. Give the name of the following simple binary ionic compounds.

- Na_2O
- K_2S
- MgCl_2
- CaBr_2
- BaI_2
- Al_2S_3
- CaBr
- AgF

2. Give the name of the following simple binary ionic compounds.

- Na_3N
- K_2O
- AgBr
- MgI_2
- SrO

3. Write the formula for the following binary ionic compounds.

- lithium bromide
- sodium iodide
- silver sulfide
- cesium oxide
- beryllium iodide
- barium hydride
- aluminum fluoride
- potassium oxide

4. Write the formula for these ionic substances.

- silver oxide
- aluminum sulfide
- sodium nitride
- barium chloride
- strontium hydride

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Table of Contents Modeling Chemistry Unit 1packet

1. Understanding the eBook Modeling Chemistry Unit 1packet
 - The Rise of Digital Reading Modeling Chemistry Unit 1packet
 - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry Unit 1packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modeling Chemistry Unit 1packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry Unit 1packet
 - Personalized Recommendations
 - Modeling Chemistry Unit 1packet User Reviews and Ratings
 - Modeling Chemistry Unit 1packet and Bestseller Lists

5. Accessing Modeling Chemistry Unit 1packet Free and Paid eBooks
 - Modeling Chemistry Unit 1packet Public Domain eBooks
 - Modeling Chemistry Unit 1packet eBook Subscription Services
 - Modeling Chemistry Unit 1packet Budget-Friendly Options
6. Navigating Modeling Chemistry Unit 1packet eBook Formats
 - ePub, PDF, MOBI, and More
 - Modeling Chemistry Unit 1packet Compatibility with Devices
 - Modeling Chemistry Unit 1packet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modeling Chemistry Unit 1packet
 - Highlighting and Note-Taking Modeling Chemistry Unit 1packet
 - Interactive Elements Modeling Chemistry Unit 1packet
8. Staying Engaged with Modeling Chemistry Unit 1packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Chemistry Unit 1packet
9. Balancing eBooks and Physical Books Modeling Chemistry Unit 1packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Chemistry Unit 1packet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modeling Chemistry Unit 1packet
 - Setting Reading Goals Modeling Chemistry Unit 1packet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modeling Chemistry Unit 1packet
 - Fact-Checking eBook Content of Modeling Chemistry Unit 1packet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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