

Chemistry – Unit 5 Worksheet 2

1. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

$$3.3 \text{ g} \times \frac{1 \text{ mole}}{63.5 \text{ g}} = 5.2 \times 10^{-2} \text{ mol or } 0.052 \text{ mol}$$

2. Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?

$$4.42 \text{ g} \times \frac{1 \text{ mole}}{55.8 \text{ g}} = 7.92 \times 10^{-2} \text{ mol or } 0.0792 \text{ mol}$$

3. A raindrop has a mass of 0.050 g. How many moles of water does a raindrop contain?

$$0.050 \text{ g} \times \frac{1 \text{ mole}}{18.0 \text{ g}} = 2.78 \times 10^{-3} \text{ mol} \rightarrow 2.8 \times 10^{-3} \text{ mol}$$

4. What mass of water would you need to have 15.0 moles of H₂O?

$$15.0 \text{ moles} \times \frac{18.0 \text{ g}}{1 \text{ mole}} = 2.70 \times 10^2 \text{ g}$$

5. One box of Morton's Salt contains 737 grams. How many moles of sodium chloride is this?

$$737 \text{ g} \times \frac{1 \text{ mole}}{58.5 \text{ g}} = 12.6 \text{ mol}$$

Na	23.0
Cl	35.5
	58.5 g

6. A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). How many grams should the chef mass out?

$$0.050 \text{ moles} \times \frac{84.0 \text{ g}}{1 \text{ mole}} = 4.2 \text{ g}$$

Na	23.0
H	1.0
C	12.0
O	48.0
	84.0

7. Rust is iron(III) oxide. The owner of a 1959 Cadillac convertible wants to restore it by removing the rust with oxalic acid, but he needs to know how many moles of rust will be involved in the reaction. How many moles of iron(III) oxide are contained in 2.50 kg of rust?

$$2.50 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ mole}}{159.6 \text{ g}} = 15.7 \text{ mol Fe}_2\text{O}_3$$

Fe	111.6
O	48.0
	159.6

Modeling Chemistry Unit 6 5

New York (State). Legislature

Modeling Chemistry Unit 6 5:

Chemistry in the Community American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design *Computational Models for Turbulent Reacting Flows* Rodney O.

Fox, 2003-10-30 Table of contents **Improving the Understanding of PM_{2.5} and Ozone Chemistry from Air Quality**

Monitoring for More Accurate Prediction of Power Generation Impacts Nancy Brown, Ling Jin, Lawrence Berkeley National Laboratory, 2013 Ultra High Pressure Treatment of Foods Marc E.G. Hendrickx, Dietrich Knorr, 2012-12-06

During the past decade consumer demand for convenient fresh like safe high quality food products has grown The food industry has responded by applying a number of new technologies including high hydrostatic pressure for food processing and preservation In addition food scientists have demonstrated the feasibility of industrial scale high pressure processing This technology is of specific interest to the food industry because it provides an attractive alternative to conventional methods of thermal processing which often produce undesirable changes in foods and hamper the balance between high quality color flavor and functionality and safety In addition it offers opportunities for creating new ingredients and products because of the specific actions of high pressure on biological materials and food constituents It allows food scientists to redesign existing processes and to create entirely new ones using high pressure technology alone or in combination with conventional processes e.g. pressure temperature combinations Researchers have investigated high pressure processing for the past century Scientists such as Hite and Bridgman did pioneering work at the turn of the 20th century Then during the 1980s and 1990s there was a large effort to investigate the effects of high pressure on biological materials particularly foods The initial research activities in the late 1980s and early 1990s focused on exploratory activities in the food area A Study of Urethane Segmented Block Copolymers with Liquid Crystalline Chain Extenders Diana Jean Gerbi, 1991

Environmental Toxicology and Chemistry, 2005 **Review of the U.S. Climate Change Science Program's Draft Synthesis and Assessment Product 2.4** National Research Council, Division on Earth and Life Studies, Board on Atmospheric Sciences and Climate, Committee to Review the U.S. Climate Change Science Program's Draft Synthesis and Assessment Product 2.4, 2007-11-30 The U.S. Climate Change Science Program is in the process of producing 21 draft assessments that investigate changes in the Earth's climate and related systems These assessments are designed to inform decisionmakers about the scientific underpinnings of a range of environmental issues such as stratospheric ozone This National Research Council report reviews one of these assessments Synthesis and Assessment Product SAP 2.4 Trends in Emissions of Ozone Depletion Substances Ozone Layer Recovery and Implications for Ultraviolet Radiation Exposure This assessment is noted as being the first ever attempt to look at the United States contribution to ozone depleting substances and ozone recovery This National Research Council book commends the assessment's authoring team for comprehensively

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Table of Contents Modeling Chemistry Unit 6 5

1. Understanding the eBook Modeling Chemistry Unit 6 5
 - The Rise of Digital Reading Modeling Chemistry Unit 6 5
 - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry Unit 6 5
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modeling Chemistry Unit 6 5
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry Unit 6 5
 - Personalized Recommendations
 - Modeling Chemistry Unit 6 5 User Reviews and Ratings
 - Modeling Chemistry Unit 6 5 and Bestseller Lists

5. Accessing Modeling Chemistry Unit 6 5 Free and Paid eBooks
 - Modeling Chemistry Unit 6 5 Public Domain eBooks
 - Modeling Chemistry Unit 6 5 eBook Subscription Services
 - Modeling Chemistry Unit 6 5 Budget-Friendly Options
6. Navigating Modeling Chemistry Unit 6 5 eBook Formats
 - ePub, PDF, MOBI, and More
 - Modeling Chemistry Unit 6 5 Compatibility with Devices
 - Modeling Chemistry Unit 6 5 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modeling Chemistry Unit 6 5
 - Highlighting and Note-Taking Modeling Chemistry Unit 6 5
 - Interactive Elements Modeling Chemistry Unit 6 5
8. Staying Engaged with Modeling Chemistry Unit 6 5
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Chemistry Unit 6 5
9. Balancing eBooks and Physical Books Modeling Chemistry Unit 6 5
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Chemistry Unit 6 5
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modeling Chemistry Unit 6 5
 - Setting Reading Goals Modeling Chemistry Unit 6 5
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modeling Chemistry Unit 6 5
 - Fact-Checking eBook Content of Modeling Chemistry Unit 6 5
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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