



Reaction Rates and Equilibria Study Guide

Topics Covered:

- Factors affecting reaction rates
- Collision theory of reaction rates
- Reaction progress diagrams (activation energy and transition states)
- Rate laws and the method of initial rates
- Chemical equilibrium
- Equilibrium expressions and constants and their meaning
- Product-favored vs. Reactant-favored
- Shifting equilibrium and Le Chatelier's Principle
- Quantitative calculations of equilibrium systems using ICE tables

Practice Questions: [Click here for an answer key](#)

- Describe how each of the following factors would affect the rate of "particle collisions" in your response.
 - Temperature decrease
 - Addition of iodide ions, I^- , as a catalyst
 - Increased reactant concentration
 - Crushing a solid into a powder (with larger surface area)
- The reaction progress diagram shows the energy changes that take nitrogen dioxide gas.



Evaluation Reaction Rates And Chemistry Equilibrium Answers

H Kauffman



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