

REVISION EXAMINATION II FOR THE ACADEMIC YEAR 2023-24

OBJECT: SCIENCE  
DATE: 07.02.2024

CLASS: X

TIME: 3 HOURS  
MAX. MARKS: 80

GENERAL INSTRUCTIONS:

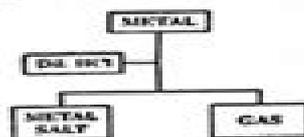
- This question paper consists of 39 questions in 5 sections.
- All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- Section A consists of 20 Objective type questions carrying 1 mark each.
- Section B consists of 6 Very Short Answer type questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- Section C consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- Section D consists of 3 Long Answer type questions carrying 05 marks each. Answers to these questions should be in the range of 80 to 120 words.
- Section E consists of 3 Source-based/Case-based units of assessment of 04 marks each with sub-parts.

SECTION A

Select and write the most appropriate option out of the four options given for each of the questions 1 - 20. There is no negative mark for incorrect response.

1. Galvanisation is a method of protecting iron from rusting by coating with a thin layer of: 1
- (a) aluminium (b) gallium  
(c) silver (d) zinc

2. Study the given flow chart carefully: 1



Which of the following two combinations are correct?

|       | Metal     | Gas Evolved |
|-------|-----------|-------------|
| (i)   | Copper    | Yes         |
| (ii)  | Iron      | Yes         |
| (iii) | Magnesium | No          |
| (iv)  | Zinc      | Yes         |

- (a) (i) and (iii) (b) (i) and (v)  
(c) (ii) and (iii) (d) (ii) and (iv)
3. Equal volumes of hydrochloric acid and sodium hydroxide solutions of same concentration are mixed and the pH of the resulting solution is checked with a pH paper. What would be the colour obtained? 1
- (a) Red (b) Blue  
(c) Orange (d) Yellowish green
4. Ankit observed that the stain of curry on a white shirt becomes reddish-brown when soap is scrubbed on it, but it turns yellow again when the shirt is washed with plenty of water. What might be the reason for his observation? 1
- (i) Soap is acidic in nature.  
(ii) Soap is basic in nature.  
(iii) Turmeric is a natural indicator which gives reddish tinge in bases.  
(iv) Turmeric is a natural indicator which gives reddish tinge in acids.
- (a) (i) and (iii) (b) (i) and (i)  
(c) (ii) and (iv) (d) (i) and (v)

# Grade 1 physical Science Question Paper 23

**Judd E. Hollander**



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