

EOC Review Packet

100%

Directions: The following questions are comprehensive review questions for the EOC exam. Review each of the items on your notes. Use the information to answer the questions on the packet.

1. Write each word using the correct number of significant figures.

a. 1.234×10^5 **1.234**
b. 5.678×10^3 **5.678**
c. 9.012×10^2 **9.012**

2. Round each number to the indicated number of significant figures.

a. 123.456 to 3 sig figs **123**
b. 0.0045678 to 2 sig figs **0.0046**
c. 1234.567 to 4 sig figs **1235**

3. Calculate the following, showing all work and using the correct number of significant figures.

a. $12.34 + 5.678$ **18.018**
b. $123.45 - 6.789$ **116.661**
c. 12.34×5.678 **70.07052**
d. $123.45 \div 6.789$ **18.184**

4. Calculate the following, showing all work and using the correct number of significant figures.

a. $12.34 + 5.678 + 9.012$ **26.03**
b. $123.45 - 6.789 - 1.234$ **115.427**
c. $12.34 \times 5.678 \times 9.012$ **624.7**
d. $123.45 \div 6.789 \div 1.234$ **14.8**

5. The following reactions are balanced. Write the balanced equation for each reaction.

a. $2H_2 + O_2 \rightarrow 2H_2O$
b. $2KClO_3 \rightarrow 2KCl + 3O_2$
c. $2H_2O \rightarrow 2H_2 + O_2$

6. Write a balanced equation for the reaction.

a. $2H_2 + O_2 \rightarrow 2H_2O$
b. $2KClO_3 \rightarrow 2KCl + 3O_2$
c. $2H_2O \rightarrow 2H_2 + O_2$

7. Calculate the molar mass of each compound.

a. H_2O **18.015 g/mol**
b. CO_2 **44.009 g/mol**
c. $NaCl$ **58.442 g/mol**
d. $CaCO_3$ **100.086 g/mol**

8. Calculate the percent composition of each compound.

a. H_2O **H: 11.1%, O: 88.9%**
b. CO_2 **C: 27.3%, O: 72.7%**
c. $NaCl$ **Na: 39.3%, Cl: 60.7%**
d. $CaCO_3$ **Ca: 28.0%, C: 12.0%, O: 60.0%**

Math Eoc Released Packet Answer

Mike Jess



Math Eoc Released Packet Answer:

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