

# Summary Sheet - Functional Group (1)

## What Are Functional Groups?

The functional groups of organic molecules are the atoms or groups of atoms that are responsible for the characteristic chemical and physical properties of the molecule.

**Alcohol** R-OH **Alcohols are functional groups that contain a hydroxyl group.**

**Structure:** R-OH



**Characteristics:** Alcohols are polar, hydrogen-bonding, and have a high boiling point. They are also soluble in water.

**Alkene** R2C=CR2 **Alkenes are functional groups that contain a carbon-carbon double bond.**

**Structure:** R2C=CR2



**Characteristics:** Alkenes are non-polar, non-hydrogen-bonding, and have a lower boiling point than alcohols. They are also soluble in water.

**Alkyne** R-C#C-R **Alkynes are functional groups that contain a carbon-carbon triple bond.**

**Structure:** R-C#C-R



**Characteristics:** Alkynes are non-polar, non-hydrogen-bonding, and have a lower boiling point than alcohols. They are also soluble in water.

**Aromatic** C6H6 **Aromatic compounds are functional groups that contain a benzene ring.**

**Structure:** C6H6



**Characteristics:** Aromatic compounds are non-polar, non-hydrogen-bonding, and have a lower boiling point than alcohols. They are also soluble in water.

**Aldehyde** R-CHO **Aldehydes are functional groups that contain a carbonyl group (C=O) and a hydrogen atom attached to the carbonyl carbon.**

**Structure:** R-CHO



**Characteristics:** Aldehydes are polar, hydrogen-bonding, and have a higher boiling point than alcohols. They are also soluble in water.

**Ketone** R2C=O **Ketones are functional groups that contain a carbonyl group (C=O) and two alkyl groups attached to the carbonyl carbon.**

**Structure:** R2C=O



**Characteristics:** Ketones are polar, hydrogen-bonding, and have a higher boiling point than alcohols. They are also soluble in water.

**Alkyne** R-C#C-R **Alkynes are functional groups that contain a carbon-carbon triple bond.**

**Structure:** R-C#C-R



**Characteristics:** Alkynes are non-polar, non-hydrogen-bonding, and have a lower boiling point than alcohols. They are also soluble in water.

**Alkyne** R-C#C-R **Alkynes are functional groups that contain a carbon-carbon triple bond.**

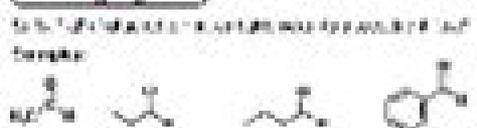
**Structure:** R-C#C-R



**Characteristics:** Alkynes are non-polar, non-hydrogen-bonding, and have a lower boiling point than alcohols. They are also soluble in water.

**Aldehyde** R-CHO **Aldehydes are functional groups that contain a carbonyl group (C=O) and a hydrogen atom attached to the carbonyl carbon.**

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**Alkyne** R-C#C-R **Alkynes are functional groups that contain a carbon-carbon triple bond.**

**Structure:** R-C#C-R



**Characteristics:** Alkynes are non-polar, non-hydrogen-bonding, and have a lower boiling point than alcohols. They are also soluble in water.

**Alcohols, Aldehydes, Ketones**

Alcohols, aldehydes, and ketones are all polar, hydrogen-bonding, and have a higher boiling point than alcohols. They are also soluble in water.

# Mastering Chemistry Answers Functional Groups

**EW Minium**



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