

5 Paragraph Biography Essay Outline

- I. Early Life (Introduction)
 - a. Birthplace and date of birth
 - b. Family background and childhood
 - c. Schools, colleges and early young adult experiences

- II. Adult Life (Body)
 - a. Marriage, children, personal life

- III. Famous years (Body)
 - a. What is this person's talent?
 - b. What makes this person stand out?
 - c. Important Accomplishments

- IV. Famous years (Body)
 - a. Why is this person one of the 100 most influential people of the year, according to Time magazine?
 - b. Did they make the world a better place, or maybe a worse place?
 - c. Did they affect people positively or negatively?

- V. Conclusion (Summary)
 - a. Give your opinion of them based on what you have learned doing your research?
 - b. Summarize your information

Cover Page

- a. The title of your paper
- b. Your name
- c. Today's date
- d. Picture of the person

Example 5 Paragraph Informative Essay Biography

Jeffrey Brautigam



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web the electron removed during the ionization of beryllium he 2s 2 is an s electron whereas the electron removed during the ionization of boron he 2s 2 2p 1 is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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potassium element has 19 electrons its electron configuration is $1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$ or simply put $4s^1$ what is the total number of electrons that can be filled in s p and d subshells

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web by building up from hydrogen this table can be used to determine the electron configuration for any atom on the periodic table we will now construct the ground state electron configuration and orbital diagram for a selection of atoms in the first and second periods of the periodic table

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