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ORANGE	GREEN	BLUE	RED	GOLD
INTRO 1-10	INTRO 1 - 10			
A 11 - 30	A 11 - 20			
B 31 - 50	B 21 - 30			
C 51 - 70	C 31 - 40	C 1 - 10		
	D 41 - 50	D 11 - 20		
	E 51 - 60	E 21 - 30		
	F 61 - 70	F 31 - 40		
	G 71 - 80	G 41 - 50		
	H 81 - 90	H 51 - 60		
	I 91 - 100	I 61 - 70		
	J 101 - 110	J 71 - 80		
		K 81 - 90		
		L 91 - 100	L 1 - 32	
		M 101 - 110	M 33 - 64	
		N 111 - 120	N 65 - 96	
			O 97 - 128	O 1 - 32
			P 129 - 160	P 33 - 64
			Q 161 - 192	Q 65 - 96
				R 97 - 128
				S 129 - 160
				T 161 - 192

	KG	FIRST	SECOND	THIRD	FOURTH	FIFTH
SEPT		D-E	I-J-K	M	P-Q	S
OCT		E-F	I-J-K	M-N	P-Q-R	S-T
NOV	A-B	F-G	J-K-L	N	Q-R	T
DEC	A-B	F-G	J-K-L	N	Q-R	T
JAN	B-C	G-H	K-L	N-O	Q-R-S	T-U
FEB	C	G-H-I	K-L-M	O	R-S	T-U
MAR	C	H-I-J	K-L-M	O-P	R-S	T-U
APR	C-D	I-J-K	L-M	O-P	R-S	U-V
MAY	D-E	I-J-K	M	P	S	U-V
JUN	E	I-J-K	M	P	S	U-V

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# Fluency Guide For Grade Levels

**Brendan G. Carr**



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$$\text{Pb(NO}_3)_2(\text{aq}) + 2\text{HCl}(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2\text{HNO}_3$$

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total ionic  $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s})$

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