

MEASURING DENSITY

READING COMPREHENSION WORKSHEET

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|---|---|---|--|
| <p>10. The density of water has a value of 1.00 g/mL.</p> <p>Complete each sentence by filling in the missing word.</p> <p>11. The density of water has a value of 1.00 g/mL.</p> <p>12. The density of water has a value of 1.00 g/mL.</p> <p>13. The density of water has a value of 1.00 g/mL.</p> <p>14. The density of water has a value of 1.00 g/mL.</p> <p>15. The density of water has a value of 1.00 g/mL.</p> <p>16. The density of water has a value of 1.00 g/mL.</p> <p>17. The density of water has a value of 1.00 g/mL.</p> <p>18. The density of water has a value of 1.00 g/mL.</p> <p>19. The density of water has a value of 1.00 g/mL.</p> <p>20. The density of water has a value of 1.00 g/mL.</p> | <p>11. The density of water has a value of 1.00 g/mL.</p> <p>12. The density of water has a value of 1.00 g/mL.</p> <p>13. The density of water has a value of 1.00 g/mL.</p> <p>14. The density of water has a value of 1.00 g/mL.</p> <p>15. The density of water has a value of 1.00 g/mL.</p> <p>16. The density of water has a value of 1.00 g/mL.</p> <p>17. The density of water has a value of 1.00 g/mL.</p> <p>18. The density of water has a value of 1.00 g/mL.</p> <p>19. The density of water has a value of 1.00 g/mL.</p> <p>20. The density of water has a value of 1.00 g/mL.</p> | <p>11. The density of water has a value of 1.00 g/mL.</p> <p>12. The density of water has a value of 1.00 g/mL.</p> <p>13. The density of water has a value of 1.00 g/mL.</p> <p>14. The density of water has a value of 1.00 g/mL.</p> <p>15. The density of water has a value of 1.00 g/mL.</p> <p>16. The density of water has a value of 1.00 g/mL.</p> <p>17. The density of water has a value of 1.00 g/mL.</p> <p>18. The density of water has a value of 1.00 g/mL.</p> <p>19. The density of water has a value of 1.00 g/mL.</p> <p>20. The density of water has a value of 1.00 g/mL.</p> | <h3>Measuring Density</h3> <p>Read the passage carefully before answering the questions that follow.</p> <p>Density is a fundamental property of matter that describes how much mass is contained in a given volume. It is calculated by dividing the mass of an object by its volume. Density is a scalar quantity, meaning it has a magnitude but no direction. It is an intensive property, meaning it does not depend on the amount of matter present. Density is a useful property for identifying substances and for understanding the behavior of matter in different states. For example, the density of water is 1.00 g/mL at 4°C. The density of air is approximately 1.29 g/L at sea level. The density of gold is 19.3 g/cm³. Density is also used to determine the purity of a substance. For example, if a sample of gold has a density of 19.3 g/cm³, it is pure gold. If the density is lower, it is an alloy. Density is also used in many other applications, such as in the design of ships and submarines. Ships are designed to float because their overall density is less than that of water. Submarines are designed to sink because their overall density is greater than that of water. Density is a key concept in science and engineering.</p> <p>Choose the option that correctly answers the question.</p> <p>1. What is the unit of density?</p> <p>2. What is the density of water at 4°C?</p> <p>3. What is the density of air at sea level?</p> <p>4. What is the density of gold?</p> |
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