

Concept Review Section 8.2

How Substances Dissolve

1. Explain how you can speed up the dissolving process when preparing a hot beverage.

2. Explain why water is sometimes referred to as the universal solvent.

3. Describe a solid substance that has particles whose interactions would be most likely to change and individual change.

4. The volume of water dissolving the same amount of sugar, salt, and potassium nitrate is added to the salt, sugar, and potassium nitrate. The three mixtures are: a. identify the particles responsible for the change.

- b. Compare the strength of the intermolecular forces between water molecules and particles of sugar, potassium nitrate, and sodium chloride.

5. Describe what happens when sodium chloride dissolves in water.

How Substances Dissolve Concept Review Answers

P. W. Bedford



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