

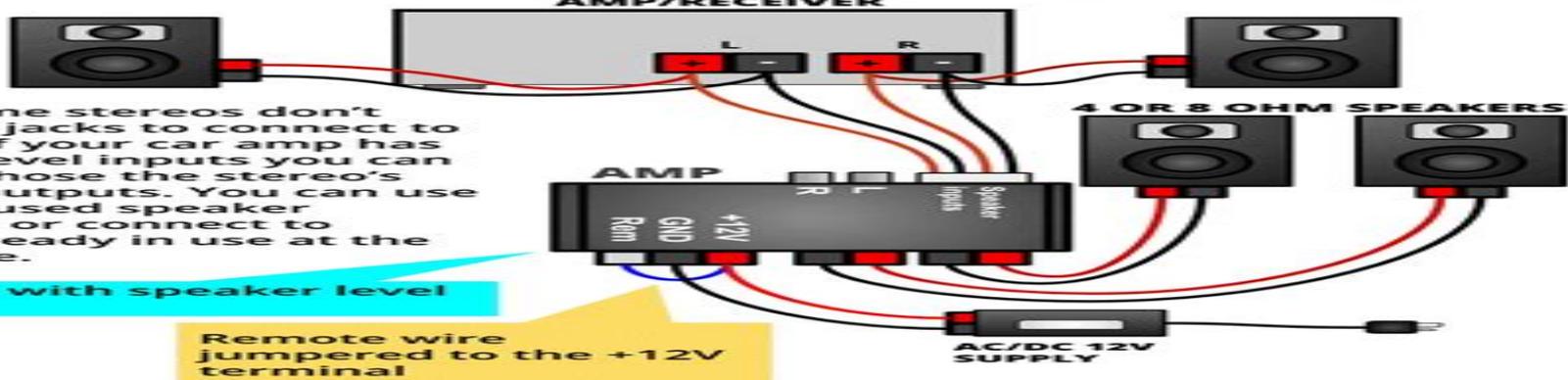
# How to connect a car amp to a home stereo

SoundCertified.com

You can connect a car amp to any (yes, any!) home stereo receiver or stereo system. The steps & parts needed depend on your car amp's features and if your home stereo has RCA jacks or only speaker outputs.

## Example 1

Home stereo with no RCA jacks + car amp with speaker level inputs



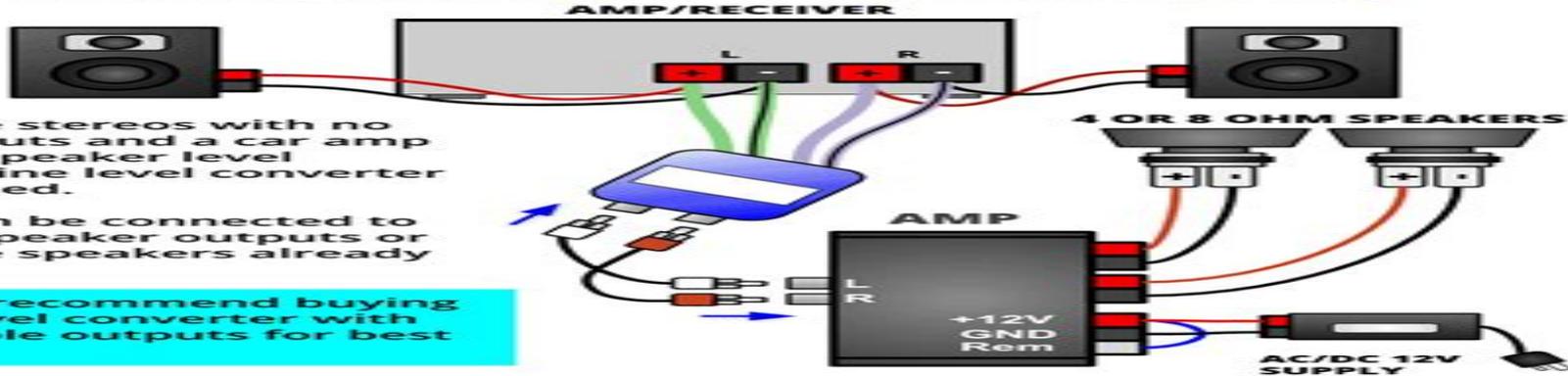
Many home stereos don't have RCA jacks to connect to an amp. If your car amp has speaker level inputs you can connect those the stereo's speaker outputs. You can use either unused speaker terminals or connect to others already in use at the same time.

Car amp with speaker level inputs

Remote wire jumpered to the +12V terminal

## Example 2

Home stereo with no RCA jacks + car amp with RCA inputs only



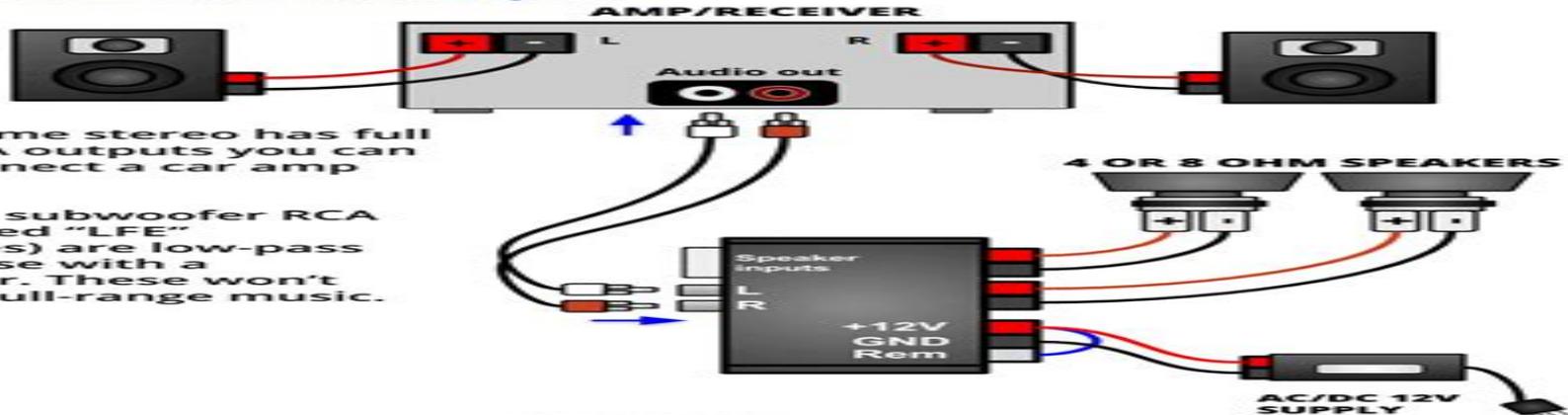
For home stereos with no RCA outputs and a car amp without speaker level inputs a line level converter can be used.

These can be connected to unused speaker outputs or alongside speakers already in use.

**NOTE:** I recommend buying a line level converter with adjustable outputs for best results.

## Example 3

Home stereo with RCA jacks



If your home stereo has full range RCA outputs you can easily connect a car amp directly.

Note that subwoofer RCA jacks (called "LFE" sometimes) are low-pass only for use with a subwoofer. These won't work for full-range music.

# How To Connect Car Amp

**M Carnoy**



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arrhenius model of bases a substance that contains a hydroxide group and dissociates to produce a hydroxide

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web kw the product of the concentrations of the hydrogen ions and hydroxide ions acidic solution one in which  $\text{H}^+$  is greater than the  $\text{OH}^-$  basic solution one in which the  $\text{H}^+$  is less than the  $\text{OH}^-$  alkaline solutions the  $\text{H}^+$  of a basic solution is less than  $1 \times 10^{-7} \text{ M}$  they are the same as basic solutions

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web jan 10 2021 to identify the strongest base we can determine their weakest conjugate acid the conjugate acids of  $\text{CH}_3\text{NH}_2$  and  $\text{S}^{2-}$  are  $\text{CH}_4$   $\text{NH}_3$  and  $\text{HS}^-$  respectively next we consider that acidity increases with positive charge on the molecule

thus ruling out that  $\text{S}^{2-}$  is the weakest base

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web compounds formed from combinations of acids and bases taste salty and are referred to in chemistry as salts sweet

compounds have characteristics of both acids and bases in the same molecule we will explore the relationship between molecular structure and acids bases and consider water solutions of acids and bases

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web conjugate acid base pair two substances related by the loss or gain of a single hydrogen ion hydronium ion  $\text{H}_3\text{O}^+$  a water molecule that gains a hydrogen ion becomes this which is positively charged what did lewis propose an acid accepts a pair of

electrons during a reaction while a base donates a pair of electrons

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web solutions manual acids i and bases section 18 1 introduction to acids and bases pages 634 643 practice problems pages 635 640 write balanced equations for reactions between the following aluminum and sulfuric acid  $2\text{Al} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{H}_2$  calcium carbonate and hydrobromic acid