

Manometer Example



$$P_{\text{atm}} = 101 \text{ kPa}$$

$$P_{\text{gas}} = P_{\text{atm}} \pm \rho g h$$

$$= 101 \text{ kPa} - (13,550 \frac{\text{kg}}{\text{m}^3}) (9.81 \frac{\text{m}}{\text{s}^2}) (0.3 \text{ m})$$

$$= 101 \text{ kPa} - 19.9 \text{ kPa}$$

$$P_{\text{gas}} = 81.1 \text{ kPa}$$

Manometer Various Problems Examples With Answers

Eugene Braunwald, Norman Hollenberg

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