



AP U.S. Government & Politics Unit 4

American Political Ideologies and Beliefs

Core American Values		Political Attitudes and Socialization	
Individualism Each person is responsible for herself, free to do what she chooses to do	Political socialization The process of a person obtaining their political ideology: family/parents, peers, education, media, religion all sources		
Equality of opportunity Each person has an opportunity to succeed, however equality of outcome is not guaranteed	Globalization U.S. political culture has both influenced and been influenced by the values of other countries		
Free enterprise An economic system based mostly on markets and freedom of people to choose what to buy and sell	Generational effects Different voting patterns and political beliefs for people in different generations		
Rule of law No person is above the law	Lifecycle effects People focus on different issues at different points in life		
Limited government Governmental power is limited by the Constitution	Major political events That occur especially during young adulthood influence political attitudes and beliefs		
Public Opinion Polling			
Types of polls		Elements of a scientific poll	
Opinion poll Poll to measure public opinion on a particular issue	Random sample All persons in the population have an equal chance being selected		
Benchmark poll Used to find out where a candidate stands before any campaigning, learn strengths, weaknesses, what issues to focus on	Stratified sample Population is divided into subgroups and weighted based on population demographics		
Tracking poll A continuous poll to chart changes in opinion over time	Sampling error A polling error arising from using only a sample of a population; +/- 3% is an acceptable margin of error		
Entrance/exit poll Taken as people enter/exit polling places on election day; used to predict election outcome, to gain insight into voter behavior, and to analyze how different demographic groups voted	Wording of question Must be neutral/unbiased, can't lead respondents towards a certain answer		
	Type and format of question Open/ended, multiple choice, answer choices, etc., can affect the poll results		
American Political Ideologies			
	Conservative	Liberal	Libertarian
Marketplace	Less economic regulation	More economic regulation	Little or no regulation beyond protection of property rights
Taxation	Lower taxes	Higher taxes on those with higher incomes	Minimal taxation

Government Unit 4 Study Guide Answers

Christian G. Meyer



Government Unit 4 Study Guide Answers:

CliffsNotes AP U.S. Government and Politics Cram Plan Jeri A. Jones, Lindsay Reeves, 2019-02-26 CliffsNotes AP U S Government Cram Plan gives you a study plan leading up to your AP exam no matter if you have two months one month or even one week left to review before the exam This new edition of CliffsNotes AP U S Government and Politics Cram Plan calendarizes a study plan for the 214 000 annual AP U S Government test takers depending on how much time they have left before they take the May exam Features of this plan to ace the exam product include 2 month study calendar and 1 month study calendar Diagnostic exam that helps test takers pinpoint strengths and weaknesses Subject reviews that include test tips and chapter end quizzes Full length model practice exam with answers and explanations [Wiley CPA Exam Review Fast Track Study Guide](#) O. Ray Whittington, 2010-06-15 THE CPA STUDY GUIDE THAT PUTS YOU ON THE FAST TRACK TO SUCCESS This tightly focused study guide now thoroughly updated to cover the latest changes in the CPA Exam cuts away all the excess information to simply give you what you need to know Presenting the absolute key points and precise steps to take in preparing for the CPA Exam the Wiley CPA Exam Review Fast Track Study Guide Fourth Edition includes sample questions and solutions that will help you master the CPA Examination O Ray Whittington well known author and professor has compiled this book using his extensive experience in preparing candidates for the CPA Exam His time honored and highly successful outline approach includes The most essential and current material needed to successfully pass the exam including discussion of the simulation style questions Thoroughly revised content to prepare readers for the new computerized CPA Exam Test taking strategies tips and techniques Coverage of new financial accounting standards including SFAS 155 Accounting for Certain Hybrid Financial Instruments 157 Fair Value Measurements and 159 Fair Value Option for Financial Assets and Financial Liabilities Coverage of new auditing standards especially AICPA Statement on Auditing Standards Nos 104 through 111 Risk Assessment Standards and the latest PCAOBs You can pass the CPA Exam Wiley CPA Exam Review Fast Track Study Guide Fourth Edition helps you prepare for master the concepts in and pass the CPA Exam Don t miss these other important study aids from the Wiley CPA Examination Review Wiley CPA Exam Review Two Volume Set VOLUME 1 Outlines and Study Guides VOLUME 2 Problems and Solutions Wiley CPA Exam Review Four Volume Set Auditing and Attestation Business Environment and Concepts Financial Accounting and Reporting Regulation Wiley CPA Exam Review Practice Software Auditing and Attestation Business Environment and Concepts Financial Accounting and Reporting Regulation Wiley CPA Exam Review Focus Notes Fifth Edition Wiley CPA Exam Review How to Master Simulations Mastering Accounting Research for the CPA Exam Second Edition Wiley CPA Exam Review Impact Audios Third Edition Auditing and Attestation Business Environment and Concepts Financial Accounting and Reporting Regulation *American Government: Annotated teacher's ed*, 1987 **Life and Liberty** Philip Roden, 1987 **Study Guide to Accompany Samuelson-Nordhaus Economics** Gary W. Yohe, 1992 **Resources in Education**, 1992-04 **An Experiment in the**

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web the coefficient in a balanced equation is an idea the concept of telling the chemist that if the atoms molecules and compounds are balanced there are balanced amounts of the atoms molecules and compounds on the other side in the product

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web the first step to balance the equation is to write down the chemical formula of reactants that are listed on the left side of the chemical equation after this you can list down the products on the right hand side of the chemical equation

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 $1\text{ n}_2 + 3\text{ h}_2 \rightarrow 2\text{ nh}_3$
 $2\text{ kclo}_3 \rightarrow 2\text{ kcl} + 3\text{ o}_2$
 $2\text{ nacl} + \text{ f}_2 \rightarrow 2\text{ naf} + \text{ cl}_2$
 $2\text{ h}_2 + \text{ o}_2 \rightarrow 2\text{ h}_2\text{ o}$
 $5\text{ pb} + 2\text{ oh} \rightarrow 2\text{ hcl} + 2\text{ h}_2 + \text{ o}_2$
 $2\text{ pbcl}_2 + 6\text{ albr}_3 + 3\text{ k}_2\text{ so}_4 \rightarrow 4\text{ kbr} + 1\text{ al}_2\text{ so}_4 + 3\text{ h}_2$
 $1\text{ ch}_4 + 2\text{ o}_2 \rightarrow 1\text{ co}_2 + 2\text{ h}_2\text{ o}$
 $8\text{ c}_3\text{ h}_8 + 5\text{ o}_2 \rightarrow 3\text{ co}_2 + 4\text{ h}_2\text{ o}$

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 $1\text{ o}_2 + 3\text{ hno}_3 \rightarrow 2\text{ h}_2\text{ o} + 2\text{ o}_2 + 3\text{ h}_2\text{ o}$
 $2\text{ o}_2 + 2\text{ h}_2\text{ o} \rightarrow 2\text{ h}_2 + \text{ o}_2$
 $4\text{ rb} + 2\text{ h}_2\text{ o} \rightarrow 2\text{ rboh} + 2\text{ h}_2$
 $5\text{ cl}_2 + 2\text{ kbr} \rightarrow 2\text{ kcl} + \text{ br}_2$
 $6\text{ n}_2 + 2\text{ h}_2 + 3\text{ nh}_3 \rightarrow 7\text{ ndf} + 3\text{ ca} + \text{ nd} + \text{ caf} + 2\text{ h}_2 + 8\text{ ch}_4$
 $4\text{ h}_2 + \text{ o}_2 \rightarrow 2\text{ h}_2\text{ o}$
 $2\text{ h}_2 + \text{ o}_2 \rightarrow 2\text{ h}_2\text{ o}$
 $9\text{ c}_2\text{ h}_5 + 5\text{ oh}$

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 $\text{ n}_2 + 3\text{ h}_2 \rightarrow 2\text{ nh}_3$
 $2\text{ kclo}_3 \rightarrow 2\text{ kcl} + 3\text{ o}_2$
 $2\text{ nacl} + \text{ f}_2 \rightarrow 2\text{ naf} + \text{ cl}_2$
 $2\text{ h}_2 + \text{ o}_2 \rightarrow 2\text{ h}_2\text{ o}$
 $\text{ pb} + \text{ oh} \rightarrow 2\text{ hcl} + \text{ h}_2\text{ o}$
 $1\text{ pbcl}_2 + 3\text{ albr}_3 + 3\text{ k}_2\text{ so}_4 \rightarrow 6\text{ kbr} + 1\text{ al}_2\text{ so}_4 + 3\text{ ch}_4 + 2\text{ o}_2$
 $\text{ c}_2\text{ h}_2 + \text{ o}_2 \rightarrow \text{ co}_2 + \text{ h}_2\text{ o}$

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coefficients equal to one 1 do not need to be shown in your answers
 $2\text{ fe} + 3\text{ cl}_2 \rightarrow 2\text{ fecl}_3$
 $4\text{ fe} + 3\text{ o}_2 \rightarrow 2\text{ fe}_2\text{ o}_3$
 $2\text{ febr}_3 + 3\text{ h}_2\text{ so}_4 \rightarrow 1\text{ fe}_2\text{ so}_4 + 3\text{ d}_1 + \text{ c}_4\text{ h}_6 + 3\text{ h}_2\text{ o} + 2\text{ c}_2\text{ h}_4 + \text{ o}_2$

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 $38\text{ fe} + 2\text{ o}_3 \rightarrow 3\text{ co} + 2\text{ fe}_3\text{ co}_2 + 39\text{ h}_2 + 3\text{ po}_4$
 $3\text{ bao} + 3\text{ po}_4 \rightarrow 2\text{ h}_2 + \text{ o}_2 + 40\text{ mg} + \text{ nh}_4 + \text{ po}_4 + 4\text{ mg} + 2\text{ p}_2 + \text{ o}_7$
 $2\text{ nh}_3 + \text{ h}_2 + \text{ o}_2 \rightarrow 4\text{ al} + 3\text{ h}_2 + \text{ so}_4 + \text{ al}_2 + \text{ so}_4 + 3\text{ h}_2 + 42\text{ p}_4 + \text{ o}_{10} + 3\text{ hcl} + \text{ pocl}_3$
 $3\text{ hpo}_3 + 43\text{ hno}_2 + 2\text{ h}_2 + \text{ o}_3 + 2\text{ hno}$

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web balancing equations the chemical equation described in section 4 1 is balanced meaning that equal numbers of atoms for

each element involved in the reaction are represented on the reactant and product sides this is a requirement the equation must satisfy to be consistent with the law of conservation of matter

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web jan 4 2019 a balanced chemical equation gives the number and type of atoms participating in a reaction the reactants products and direction of the reaction balancing an unbalanced equation is mostly a matter of making certain mass and charge are balanced on the reactants and products side of the reaction arrow

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