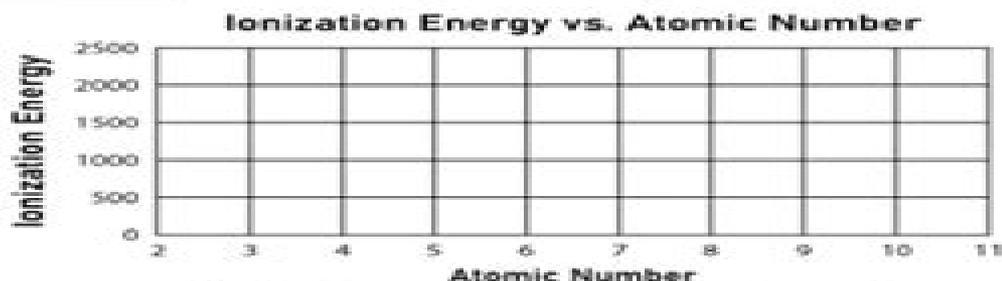


Graphing Periodic Trends Worksheet

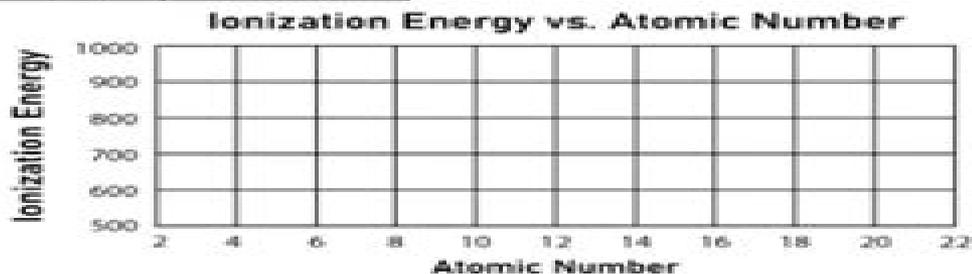
1. Using the ionization energy (IE) values of the elements in Period 2 listed below, make a line graph of IE vs. the atomic number.

Period 2					
Element	Atomic Number	IE (kJ/mol)	Element	Atomic Number	IE (kJ/mol)
Li	3	519	Li	3	519
Be	4	900	Be	4	900
B	5	799	B	5	799
C	6	1088	C	6	1088



- a) What trend do you notice for the ionization energies in Period 2?
- b) Explain why you observe this trend.
2. Make a scatter plot of ionization energy vs. atomic number for the first three atoms in Group 2A listed below.

Group 2A		
Element	Atomic Number	IE (kJ/mol)
Be	4	900
Mg	12	736
Ca	20	590



Graphing Periodic Trends Answer Sheet

BM King



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Table of Contents Graphing Periodic Trends Answer Sheet

1. Understanding the eBook Graphing Periodic Trends Answer Sheet
 - The Rise of Digital Reading Graphing Periodic Trends Answer Sheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Graphing Periodic Trends Answer Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Graphing Periodic Trends Answer Sheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Graphing Periodic Trends Answer Sheet
 - Personalized Recommendations
 - Graphing Periodic Trends Answer Sheet User Reviews and Ratings
 - Graphing Periodic Trends Answer Sheet and Bestseller Lists
5. Accessing Graphing Periodic Trends Answer Sheet Free and Paid eBooks
 - Graphing Periodic Trends Answer Sheet Public Domain eBooks
 - Graphing Periodic Trends Answer Sheet eBook Subscription Services
 - Graphing Periodic Trends Answer Sheet Budget-Friendly Options
6. Navigating Graphing Periodic Trends Answer Sheet eBook Formats

- ePub, PDF, MOBI, and More
 - Graphing Periodic Trends Answer Sheet Compatibility with Devices
 - Graphing Periodic Trends Answer Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Graphing Periodic Trends Answer Sheet
 - Highlighting and Note-Taking Graphing Periodic Trends Answer Sheet
 - Interactive Elements Graphing Periodic Trends Answer Sheet
 8. Staying Engaged with Graphing Periodic Trends Answer Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Graphing Periodic Trends Answer Sheet
 9. Balancing eBooks and Physical Books Graphing Periodic Trends Answer Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Graphing Periodic Trends Answer Sheet
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Graphing Periodic Trends Answer Sheet
 - Setting Reading Goals Graphing Periodic Trends Answer Sheet
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Graphing Periodic Trends Answer Sheet
 - Fact-Checking eBook Content of Graphing Periodic Trends Answer Sheet
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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