

Reminders:

1. In a neutral atom the number of protons equals the number of electrons.
2. An atom can NEVER gain or lose protons.
3. The number of protons equals the atomic number.

Important!

NAME: _____

Answer Key**Ion Practice Set**

1. What is an ion?

or a charged ion - it can be positively charged (cation) or negatively charged (anion).

2. What does the number next to the ions signify?

the charge of the ion

Complete the following table, using the periodic table in the back of your book.

| | ELEMENT NAME | ION SYMBOL | NUMBER OF PROTONS | NUMBER OF ELECTRONS | NUMBER OF ELECTRONS LOST OR GAINED |
|----|--------------|------------------|-------------------|---------------------|------------------------------------|
| ex | Fluorine | F ⁻ | 9 | 10 | gained one |
| 1 | Iodine | I ⁻ | 53 | 54 | gained one |
| 2 | Sulfur | S ²⁻ | 16 | 16+2 = <u>18</u> | gained two |
| 3 | potassium | K ⁺¹ | 19 | 19-1 = <u>18</u> | lost one |
| 4 | Calcium | Ca ⁺² | 20 | 20-2 = <u>18</u> | lost two |
| 5 | Bromine | Br ⁻ | 35 | 36 | gained one |
| 6 | Strontium | Sr ⁺² | 38 | 38-2 = <u>36</u> | lost two |
| 7 | Hydrogen | H ⁺ | 1 | 1-1 = <u>0</u> | lost one |
| 8 | Oxygen | O ²⁻ | 8 | 8+2 = <u>10</u> | gained two |
| 9 | Magnesium | Mg ⁺² | 12 | 12-2 = <u>10</u> | lost two |
| 10 | aluminum | Al ⁺³ | 13 | 10 | lost three |
| 11 | Selenium | Se ²⁻ | 34 | 36 | gained two |
| 12 | Hydrogen | H ⁺ | 1 | 1+1 = <u>2</u> | gained one |
| 13 | lithium | Li ⁺¹ | 3 | 3-1 = <u>2</u> | lost one |
| 14 | Rubidium | Rb ⁺ | 37 | 37-1 = <u>36</u> | lost one |
| 15 | Chlorine | Cl ⁻ | 17 | 18 | gained one |

this means it has an extra electron

this means it has 2 more protons than electrons, so 2 electrons were lost!

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RS Peters



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Table of Contents Ion Practice Set Answer Key

1. Understanding the eBook Ion Practice Set Answer Key
 - The Rise of Digital Reading Ion Practice Set Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ion Practice Set Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ion Practice Set Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ion Practice Set Answer Key
 - Personalized Recommendations
 - Ion Practice Set Answer Key User Reviews and Ratings
 - Ion Practice Set Answer Key and Bestseller Lists
5. Accessing Ion Practice Set Answer Key Free and Paid eBooks
 - Ion Practice Set Answer Key Public Domain eBooks
 - Ion Practice Set Answer Key eBook Subscription Services

- Ion Practice Set Answer Key Budget-Friendly Options
- 6. Navigating Ion Practice Set Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ion Practice Set Answer Key Compatibility with Devices
 - Ion Practice Set Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ion Practice Set Answer Key
 - Highlighting and Note-Taking Ion Practice Set Answer Key
 - Interactive Elements Ion Practice Set Answer Key
- 8. Staying Engaged with Ion Practice Set Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ion Practice Set Answer Key
- 9. Balancing eBooks and Physical Books Ion Practice Set Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ion Practice Set Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ion Practice Set Answer Key
 - Setting Reading Goals Ion Practice Set Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ion Practice Set Answer Key
 - Fact-Checking eBook Content of Ion Practice Set Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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