

INTRO TO IONIC BONDING

GUIDED NOTES



Ionic Bonding

All ions have a charge (lost electrons = positive charge, gained electrons = negative charge). Ions are described by their charge and their size.

Write down the names of the following ions. Describe how they will differ from their parent atoms.

POSITIVE IONS
Metals lose electrons. Write down the names of the following metals. How many electrons do they lose? How many protons do they have? How many neutrons do they have? How many shells do they have? How many electrons do they have? How many shells do they have? How many electrons do they have?

NEGATIVE IONS
Non-metals gain electrons. Write down the names of the following non-metals. How many electrons do they gain? How many protons do they have? How many neutrons do they have? How many shells do they have? How many electrons do they have?

IONIC BONDING
Write down the names of the following ionic compounds. How many electrons do they have? How many protons do they have? How many neutrons do they have? How many shells do they have? How many electrons do they have?

SELF ASSESSMENT QUESTIONS

Write down the names of the following ions. Describe how they will differ from their parent atoms.

Write down the names of the following ionic compounds. How many electrons do they have? How many protons do they have? How many neutrons do they have? How many shells do they have? How many electrons do they have?

IONIC BONDING

Write down the names of the following ions. Describe how they will differ from their parent atoms.

Write down the names of the following ionic compounds. How many electrons do they have? How many protons do they have? How many neutrons do they have? How many shells do they have? How many electrons do they have?

IONIC BONDING

Name: _____

Directions: Answer the following.

Write W for metal and N for non-metal for each of the following.

- _____ gains electrons when bonding
- _____ found on the left side of the periodic table
- _____ has just a few valence electrons
- _____ aluminum
- _____ has more than four valence electrons
- _____ loses electrons when bonding
- _____ found on the right side of the periodic table
- _____ bromine

Draw a Lewis structure for each of the following ionic compounds.

Strontium + Oxygen	Hydrogen + Chlorine
Hydrogen + Oxygen	Calcium + Phosphorus

Self Assessment Qs

Classwork Worksheet



Ionic Bonding Notes Guide

Padhraic Smyth



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