

Name: KEY

Matter Unit Test

Introduction To Matter Section 1 - Multiple Choice Questions are 4 points each

P 1. Matter is defined as any object that _____

- a. Has mass
- b. Has volume
- c. Is alive - *metals/salt, sugar not alive but matter*
- d. Has mass and volume (takes up space)

if it has mass, it must have volume

C 2. Which of the following statements about the periodic table is TRUE?

- a. All elements on the periodic table belong to the same family.
- b. Elements on the periodic table are organized by where they are found in nature.
- c. Elements on the periodic table are unique from each other, but do share common characteristics with other elements.
- d. Elements on the periodic table can be broken down into more simple substances.

weight

they cannot

B 3. A piece of iron and a piece of gold have the same mass. Which statement best describes the metals?

- a. The amount of matter in the iron is different than the amount of matter in the gold. *-not enough info*
- b. The number of atoms in the iron is different than the number of atoms in gold.
- c. The melting point of the iron is the same as the melting point of the gold. *different atoms*
- d. The volume of the iron will be the same as the volume of gold. *-not enough info*

B 4. If we were to describe the relationship between atoms, matter and elements, using the United States, which of the following statements is true?

Small Large Medium

- a. The whole US would be atoms, each state would be elements and the cities in that state would be matter. *-X*
- b. The whole US would be matter, each state would be elements and the cities in that state would be atoms. *E L M S*
- c. The whole US would be elements, each state would be atoms and the cities in that state would be matter.
- d. The whole US would be matter, each state would be atoms and the cities in that state would be elements.

Introduction To Matter Answer Key

TD Snyder



Introduction To Matter Answer Key:

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