

## Gas Laws Simulation

Use the following simulation to explore the relationship between pressure, volume, and temperature.

1. Describe the three different gases (pressure, number of "molecules", and mass).
2. Describe "weight".



3. For each gas, describe the relationship between pressure and volume. What is the effect?
4. How do pressure, number of particles, and mass affect volume? What is the effect?
5. What is the relationship between  $P$  and  $V$  if the other variables are held constant?
6. How do pressure, number of particles, and mass affect volume? What is the effect?

Investigate the relationship between

1. temperature and volume at the following: hold the pressure and number of molecules constant. What is the effect?
2. What is the relationship between  $P$  and  $V$ ?
3. How do the number of particles, mass, and temperature affect the volume?
4. What is the relationship between  $P$  and  $V$ ?
5. How do the number of particles, mass, and temperature affect the volume?

Temperature, constant number of molecules	
Temperature (K)	Volume (m³)
300	1.00
400	1.33
500	1.50

6. What is the relationship between temperature and volume if the gas molecules are present at constant volume? As the temperature of a quantity of the gas increases, the pressure at constant volume increases.

# Exploring Gas Laws Simulation Answers

**Michael Seilmaier**



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