

Gas Laws Simulation

Use the following simulation to explore the relationship between pressure, volume, and temperature.

1. Describe the three different gases (pressure, molecular mass/weight, and speed/velocity).



2. For each gas, describe the quantity being represented, what it would mean to hold it **constant**.
3. How do pressure, Particle Mass, and particle volume **and volume** (about 1/3rd of 1.00).
4. How do temperature and **g** affect each quantity in the **1.00**.
5. How do pressure, Particle Mass, and velocity **and moving randomly along the container** (not straight or in 1 direction).

Investigate the relationship:

1. **Temperature and Pressure** at the following, hold the quantity and temperature constant (at 1.00):
2. Hold the temperature at 1.
3. Record the measurements **at least 10 times**.
4. Hold the temperature at 2.
5. Record the measurements **at least 10 times**.

Temperature, constant volume, constant mass	
Temperature (K)	Pressure (atm)
1.000	1.000
2.000	2.000
3.000	3.000

6. **Volume and Temperature** (at constant pressure) - vary the gas temperature to pressure at constant volume. As the temperature of a quantity of the gas increases, the pressure at constant volume increases.

Exploring The Gas Laws Simulation Part Answers

Guillaume Favre



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