

Test "Chemistry"

Chapter # 3

Objective part

Time: 15 min

Total marks: 12

Q.1- Every question has four possible answers. Encircle the correct option.

- i) The atomic radii of the elements in periodic table;
a) increase from left to right in a period
b) increase from top to bottom in a group
c) do not change from left to right in a period
d) decrease from top to bottom in a group
- ii) Mendeleev Periodic table was based upon the:
a) electronic configuration
b) atomic mass
c) atomic number
d) electro negativity
- iii) The amount of energy given out when an electron is added to an atom is called;
a) ionization energy
b) lattice energy
c) electron affinity
d) electro negativity
- iv) Long form of periodic table is constructed on the basis of:
a) Mendeleev postulate
b) atomic number
c) atomic mass
d) mass number
- v) Along the period which of the following decreases:
a) electro negativity
b) ionization energy
c) atomic radius
d) electron affinity
- vi) Mark the incorrect statement about the ionization energy:
a) it is measured kJ mol^{-1}
b) it is the absorption of energy
c) it decreases in a period
d) it decreases in groups
- vii) Transition elements are:
a) all gases
b) all metals
c) all non metals
d) all metalloids
- viii) S-f series of inner transition elements are called:
a) lanthanides
b) actinides
c) halogens
d) alkali metals
- ix) If we move from left to right in a period the value of ionization energy:
a) remain same
b) decreases
c) increases
d) not effected
- x) Which scientist has given the idea of actives in periodic table:
a) Mendeleev
b) Al-razi
c) Newland
d) dobernier
- xi) Shielding effect is greater in atoms with greater number of:
a) proton
b) electron
c) neutron
d) positron
- xii) Which is the incomplete period in the periodic table:
a) 4th period
b) 2nd period
c) 7th period
d) 1st period

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C Cleary



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Modern NMR Techniques for Chemistry

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Henderson, 2017-12-21 This popular and comprehensive textbook provides all the basic information on inorganic chemistry that undergraduates need to know For this sixth edition the contents have undergone a complete revision to reflect progress in areas of research new and modified techniques and their applications and use of software packages Introduction to Modern Inorganic Chemistry begins by explaining the electronic structure and properties of atoms then describes the principles of bonding in diatomic and polyatomic covalent molecules the solid state and solution chemistry Further on in the book the general properties of the periodic table are studied along with specific elements and groups such as hydrogen the s elements the lanthanides the actinides the transition metals and the p block Simple and advanced examples are mixed

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Foreign Returns Great Britain. Royal Commission on Elementary Education Acts,1888

Uneducated Guesses Howard Wainer,2011-08-08 What statistical evidence shows us about our misguided educational policies Uneducated Guesses challenges everything our policymakers thought they knew about education and education reform from how to close the achievement gap in public schools to admission standards for top universities In this explosive book Howard Wainer uses statistical evidence to show why some of the most widely held beliefs in education today and the policies that have resulted are wrong He shows why colleges that make the SAT optional for applicants end up with underperforming students and inflated national rankings and why the push to substitute achievement tests for aptitude tests makes no sense Wainer challenges the thinking behind the enormous rise of advanced placement courses in high schools and demonstrates why assessing teachers based on how well their students perform on tests a central pillar of recent education reforms is woefully misguided He explains why college rankings are often lacking in hard evidence why essay questions on tests disadvantage women why the most grievous errors in education testing are not made by testing organizations and much more No one concerned about seeing our children achieve their full potential can afford to ignore this book With forceful storytelling wry insight and a wealth of real world examples Uneducated Guesses exposes today s educational policies to the light of empirical evidence and offers solutions for fairer and more viable future policies

Holt Science and Technology

Holt Rinehart & Winston,2004-02

Voet's Principles of Biochemistry, Global Edition Donald Voet,Judith G.

Voet,Charlotte W. Pratt,2018-01-11 Voet s Principles of Biochemistry Global Edition addresses the enormous advances in biochemistry particularly in the areas of structural biology and bioinformatics It provides a solid biochemical foundation that is rooted in chemistry to prepare students for the scientific challenges of the future New information related to advances in biochemistry and experimental approaches for studying complex systems are introduced Notes on a variety of human diseases and pharmacological effectors have been expanded to reflect recent research findings While continuing in its tradition of presenting complete and balanced coverage this Global Edition includes new pedagogy and enhanced visuals that provide a clear pathway for student learning

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