

Modern Chemistry Chapter 7

Chemical Formulas & Chemical Compounds

- A chemical formula indicates the kind and relative number of atoms in a chemical compound.
- C_8H_{18} (octane) has 8 carbon and 18 hydrogen atoms.



Forming Ionic Compounds

- Compounds that have the elements held together by ionic bonds are called ionic compounds.
- For an ionic compound to exist, the algebraic sum of the positive and the negative charges of the ions **MUST = 0**.
- For instance, when a calcium atom becomes an ion, it has an overall 2+ charge which must be neutralized by ion(s) that have a 2- charge.
- IF a Ca^{2+} cation forms an ionic bond with an O^{2-} anion, the resulting compound will be neutral and the formula would be CaO .

Modern Chemistry Chapter 7 Homework 7 10 Worksheet Answers

Kenneth T. Henson



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Books in Print Supplement, 2002 Children's Books in Print R R Bowker Publishing, Bowker, 1999-12 *Secondary Teaching Methods* Kenneth T. Henson, 1981 **Modern Chemistry** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

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