

Molarity Practice Worksheet

Find the molarity of the following solutions:

- 1) 0.5 moles of sodium chloride is dissolved to make 0.05 liters of solution.

- 2) 0.5 grams of sodium chloride is dissolved to make 0.05 liters of solution.

- 3) 0.5 grams of sodium chloride is dissolved to make 0.05 mL of solution.

- 4) 734 grams of lithium sulfate are dissolved to make 2500 mL of solution.

- 5) 6.7×10^{-2} grams of $\text{Pb}(\text{C}_2\text{H}_3\text{O}_2)_4$ are dissolved to make 3.5 mL of solution.

- 6) I have two solutions. In the first solution, 1.0 moles of sodium chloride is dissolved to make 1.0 liters of solution. In the second one, 1.0 moles of sodium chloride is added to 1.0 liters of water. Is the molarity of each solution the same? Explain your answer.

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LL Leslie



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Nursing Care Planning Guides Susan Puderbaugh Ulrich, Suzanne Weyland Canale, Sharon Andrea Wendell, 1990 This new edition provides 75 nursing care plans for the care of adult clients with common medical surgical conditions Each plan includes all relevant nursing and collaborative diagnoses in depth aetiologies measurable outcome criteria and nursing interventions **Journal** American Dietetic Association, 1948 *Comprehensive Dissertation Index, 1861-1972: Education* Xerox University Microfilms, 1973 *CIS Index to U.S. Executive Branch Documents, 1910-1932* Congressional Information Service, 1996

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