

The Mole Study Guide

The Mole

A mole is a unit of measurement for the amount of a substance. It is defined as the amount of substance that contains as many elementary entities as there are atoms in 12 grams of carbon-12.

1 mole = 6.02×10^{23} particles

Substance	Mass (g)	Number of Particles
Carbon-12	12	6.02×10^{23}
Hydrogen	1	6.02×10^{23}
Oxygen	16	6.02×10^{23}
Water	18	6.02×10^{23}

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