



PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2

TIME: 1 HOUR

VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).

1. Write a scientific report which must include the following:
- 1.1 Aim (1)
- 1.2 Apparatus and method (5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (°)

(6)

3. Analysis of the results:
- 3.1 Identify the following variables for the experiment:
- Independent
 - Dependent
 - Controlled
- (3)

- 3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$. (18)

4. Conclusion

- 4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased. (1)
- 4.2 Write down a conclusion for this experiment. (3)

5

Questions and discussion:

- 5.1 Explain why you have to wait for a few minutes before you can take readings. (2)
- 5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases. (4)

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Table of Contents Grade11 Physics Control Question Paper Of March2014

1. Understanding the eBook Grade11 Physics Control Question Paper Of March2014
 - The Rise of Digital Reading Grade11 Physics Control Question Paper Of March2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade11 Physics Control Question Paper Of March2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade11 Physics Control Question Paper Of March2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade11 Physics Control Question Paper Of March2014
 - Personalized Recommendations
 - Grade11 Physics Control Question Paper Of March2014 User Reviews and Ratings

- Grade11 Physics Control Question Paper Of March2014 and Bestseller Lists
- 5. Accessing Grade11 Physics Control Question Paper Of March2014 Free and Paid eBooks
 - Grade11 Physics Control Question Paper Of March2014 Public Domain eBooks
 - Grade11 Physics Control Question Paper Of March2014 eBook Subscription Services
 - Grade11 Physics Control Question Paper Of March2014 Budget-Friendly Options
- 6. Navigating Grade11 Physics Control Question Paper Of March2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade11 Physics Control Question Paper Of March2014 Compatibility with Devices
 - Grade11 Physics Control Question Paper Of March2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade11 Physics Control Question Paper Of March2014
 - Highlighting and Note-Taking Grade11 Physics Control Question Paper Of March2014
 - Interactive Elements Grade11 Physics Control Question Paper Of March2014
- 8. Staying Engaged with Grade11 Physics Control Question Paper Of March2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade11 Physics Control Question Paper Of March2014
- 9. Balancing eBooks and Physical Books Grade11 Physics Control Question Paper Of March2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade11 Physics Control Question Paper Of March2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade11 Physics Control Question Paper Of March2014
 - Setting Reading Goals Grade11 Physics Control Question Paper Of March2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade11 Physics Control Question Paper Of March2014
 - Fact-Checking eBook Content of Grade11 Physics Control Question Paper Of March2014
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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