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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{T} \times n_0$ 1 2 4 x 50 3 125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{T} \times n_0$ 100 now using the half life equation in example 1 we have

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isotope has a half life of 10 years

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web chapter 15 nuclear chemistry half life learning objectives define half life determine the amount of radioactive substance remaining after a given number of half lives whether or not a given isotope is radioactive is a characteristic of that particular isotope

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web the half life of a radioactive isotope is the amount of time it takes for one half of the radioactive isotope to decay the half life of a specific radioactive isotope is constant it is unaffected by conditions and is independent of the initial amount of that isotope

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web section learning objectives by the end of this section you will be able to do the following explain radioactive half life and its role in radiometric dating calculate radioactive half life and solve problems associated with radiometric dating section key terms half life and the rate of radioactive decay unstable nuclei decay

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web solution if we compare the time that has passed to the isotope s half life we note that 44 0 s is exactly 4 half lives so using the previous equation $n = 4$ substituting and solving results in the following less than one third of a gram of 20 f remains skill building exercise the half life of 44 ti is 60 0 y

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web solution 7 2 2 4 3 half lives 1 2 0 125 the amount remaining after 3 half lives 100 0 g \times 0 125 12 5 g remaining problem 2 pd 100 has a half life of 3 6 days if one had 6.02×10^{23} atoms at the start how many atoms would be present after 20 0 days solution 20 0 3 6 5 56 half lives

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each isotope used in these applications has a characteristic half life