

# Empirical Formula

The simplest formula of a compound, which represents the whole number ratio of atoms of different elements present in a compound or a class of compounds, is called the empirical formula.

Molecular Formula	Empirical Formula
$\text{CO}_2$	$\text{CO}_2$
$\text{H}_2\text{O}$	$\text{H}_2\text{O}$
$\text{N}_2\text{O}_4$	$\text{NO}_2$
$\text{C}_6\text{H}_{12}\text{O}_6$	$\text{CH}_2\text{O}$
$\text{P}_4\text{O}_{10}$	$\text{P}_2\text{O}_5$
$\text{H}_2\text{O}_2$	$\text{HO}$

# How To Calculate Empirical And Molecular Formula S

**David Baud**



## **How To Calculate Empirical And Molecular Formula S:**

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