

MYTHICAL CREATURES



Research

Watch a video about mythical creatures [here](#).
Click each image for books about mythical creatures.



Present

Create your own poster, slideshow, factsheet or video about a mythical creature to share what you've learnt.



Write

Write your own story, song or poem about mythical creatures.
Click [here](#) for some musical inspiration.



Investigate

There are lots of mythical creatures from Māori legends in New Zealand?
Click [here](#) to find out more about them.



Discover

Click the image below to find out about taniwha - mythical creatures from Māori legends.



Design

Design your own mythical creature.
Use [these prompt sheets](#) to help you gather some ideas.



Read

Here are some fun mythical stories by Holly Hatam, perfect for reading out loud:
Dragons are real!
Fairies are real!
Mermaids are real!
Unicorns are real!

Move and Play

Move like a mythical creature with a game of Simon Says...

Fly like a fairy
Stay still like sphinx
Prowl like a dragon
Stride like a yeti



Generate a Creature

Click below to open an online magical book that will summon mythical creatures!



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Key Stage2 Mythological Creatures

Patrick Vollmar



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 $\text{H}_2\text{O g} + \text{H}_2 \text{ g} + \frac{1}{2} \text{O}_2 \text{ g} \Delta h 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} + \frac{7}{2} \text{O}_2 \text{ g} \rightarrow \text{CO}_2 \text{ g} + 3 \text{H}_2\text{O g} \Delta h 283 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g}$

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$\text{C}_2\text{H}_6 \text{ g} + 7 \text{O}_2 \text{ g} \rightarrow 2 \text{CO}_2 \text{ g} + 3 \text{H}_2\text{O l} \Delta h 1560 \text{ kJ mole}$
 $\text{H}_2 \text{ g} + \frac{1}{2} \text{O}_2 \text{ g} \rightarrow \text{H}_2\text{O l} \Delta h 285 \text{ kJ mole}$

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web may 8 2015 hess law lab by maya parks partners ben seufert kelsea floyd convert joules to kj in your final answer multiply the mass by the change in temperature and the c given trial 1 use hess s law and the following equations and ΔH values to determine the heat of reaction for the reaction C_2H_4

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web thus for reaction 9 2 in which 1 mole of H_2SO_4 and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta H = 55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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web hess s law enthalpy of reaction and hess s law abstract the goal of this lab is to perform experiments and practice hess s law by determining enthalpy change in chemical reactions in this lab three experiments were conducted in order to test hess s law the experiments required observing and collecting data on three different reactions

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web objective the purpose of the lab is to combine two reactants in the calorimeter and measure the heats of reaction in solution eventually allowing an individual to prove or disprove hess s law introduction thermodynamics is the study of energy exchanged between the system and the surroundings

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web 3 12 hess s law in this experiment you will measure the amount of heat released in these three related exothermic reactions 1 $\text{NaOH}(\text{s}) + \text{Na}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l})$ 2 $\text{NaOH}(\text{s}) + \text{H}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l})$ 3