

13. Complete the following table for the three experiments in Model 2.

	Experiment C	Experiment D	Experiment E
Independent Variable	# of particles	Temp.	Internal Pressure External Pressure
Dependent Variable	Volume	Volume	Volume
Controlled Variable(s)	Temp / I+E pressure	External + Internal Pressure / # of particles	Temp / # of particles

14. Provide a molecular level explanation for the increase in volume among the balloons in Experiment C. (How often and/or how hard are the molecules hitting the sides of the container?)

# of molecules ↑, so number of collisions ↑ + Volume along with it

collisions ↑ energy stays the same

15. Provide a molecular level explanation for the increase in volume among the balloons in Experiment D.

Temperature ↑ increasing the kinetic energy of the particles as well which ↑ collisions thus ↑ pressure

16. Provide a molecular level explanation for the increase in volume among the balloons in Experiment E.

if external forces decrease then it allows internal forces ↑ thus ↑ volume

# Gas Variables Pogil Activities With Answers

**Jicheng Xie**



## **Gas Variables Pogil Activities With Answers:**

## **Gas Variables Pogil Activities With Answers** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Gas Variables Pogil Activities With Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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