

## SHORT ANSWER QUESTIONS (From Previous Board Papers 2014-2015-2016-2017)

1. Why HF is weaker acid than HCl? (LHR, GH, 2015)

Ans. HF is weaker than HCl because ionization of HF is very small (10% only). Due to strong hydrogen bonding, HF has less ionization.

2. Give reason why fluorine is a gas, iodine is solid. (GRW, 2017)

Ans. Fluorine is a gas because it has small size, low polarizability and hence weak van der Waals forces, whereas iodine has large size, greater polarizability and strong London forces which make it solid.

3. In what ways fluorine differs from its group members? (BWP, SGB, GI, 2015)(MLN, GH, 2016)(SGD, GI, 2017)

Ans. **Peculiar Behaviour of Fluorine:** Fluorine shows peculiar behaviour in many respects which is due to:

- (i) Small size of F atom and of  $F^-$  ion.                      (ii) High first ionization energy and electronegativity.
- (iii) Low dissociation energy of  $F_2$  molecule as compared to  $Cl_2$  and  $Br_2$ .
- (iv) Restriction of valence shell to an octet.              (v) Direct combination with inert gases.

4. Describe two uses of helium. (LHR, GI, 2016)

Ans. Helium is first member of group VIII A, and found as a mono atomic gas in nature. Helium has following important uses.

(i) **In weather balloons:** Helium is used in weather balloons, in welding and in traffic signal light.

(ii) **In sea diving:** A mixture of 80% helium and 20% oxygen is used for breathing by the sea divers.

(iii) **In nuclear reactors:** Helium is used as a cooling medium for nuclear reactors.

5. Halogens are strong oxidizing agents justify. (SWL, MLN, GH, 2014) (MLN, GI, 2016)

Ans. All halogens have positive values of reduction potential, they have tendency to accept electron and form negative ion. They can remove electron from other elements and oxidize them to positive ions, e.g.  $2Na + F_2 \longrightarrow 2NaF$

Hence it is justified that halogens are oxidizing agents. The order of oxidizing power of halogens is as follows:  $F_2 > Cl_2 > Br_2 > I_2$

# Gases Section Short Answers

**J Elliott**



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web 500 000 sqft 1 000 000 sqft budget 10m 50m the meydan retail complex and multiplex performs not just as an efficient

retail complex but as a true urban centre in