

**Question Number : 7****Correct : 1 Wrong : 0**

The dependence of rate constant  $k$  on temperature  $T$  (in K) of a reaction is given by the expression

$$\ln k = [(-5000 \text{ K})/T] + 10$$

The activation energy of the reaction (in  $\text{kJ mol}^{-1}$ ) is \_\_\_\_\_ (up to two decimal places)

**Question Number : 8****Correct : 1 Wrong : 0**

The lowest energy of a quantum mechanical one-dimensional simple harmonic oscillator is  $300 \text{ cm}^{-1}$ . The energy (in  $\text{cm}^{-1}$ ) of the next higher level is \_\_\_\_\_

**Question Number : 9****Correct : 1 Wrong : -0.33**

The electronic ground state term for the chromium ion in  $[\text{Cr}(\text{CN})_6]^{4-}$  is

- (A)  $^3F$                       (B)  $^3H$                       (C)  $^3G$                       (D)  $^5D$

**Question Number : 10****Correct : 1 Wrong : -0.33**

The  $\text{VO}_4^{3-}$ ,  $\text{CrO}_4^{2-}$  and  $\text{MnO}_4^-$  ions exhibit intense ligand to metal charge transfer transition. The wavelengths of this transition follow the order

- (A)  $\text{CrO}_4^{2-} < \text{VO}_4^{3-} < \text{MnO}_4^-$                       (B)  $\text{MnO}_4^- < \text{VO}_4^{3-} < \text{CrO}_4^{2-}$   
(C)  $\text{VO}_4^{3-} < \text{CrO}_4^{2-} < \text{MnO}_4^-$                       (D)  $\text{CrO}_4^{2-} < \text{MnO}_4^- < \text{VO}_4^{3-}$

**Question Number : 11****Correct : 1 Wrong : -0.33**

The lanthanide ion that exhibits color in aqueous solution is

- (A)  $\text{La}(\text{III})$                       (B)  $\text{Eu}(\text{III})$                       (C)  $\text{Gd}(\text{III})$                       (D)  $\text{Lu}(\text{III})$

**Question Number : 12****Correct : 1 Wrong : 0**

The hapticity of cycloheptatriene,  $(\text{C}_7\text{H}_8)$ , in  $\text{Mo}(\text{C}_7\text{H}_8)(\text{CO})_3$  is \_\_\_\_\_

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## **Table of Contents Gate Previous Year Question Papers For Chemistry**

1. Understanding the eBook Gate Previous Year Question Papers For Chemistry
  - The Rise of Digital Reading Gate Previous Year Question Papers For Chemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Gate Previous Year Question Papers For Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gate Previous Year Question Papers For Chemistry
  - User-Friendly Interface

4. Exploring eBook Recommendations from Gate Previous Year Question Papers For Chemistry
  - Personalized Recommendations
  - Gate Previous Year Question Papers For Chemistry User Reviews and Ratings
  - Gate Previous Year Question Papers For Chemistry and Bestseller Lists
5. Accessing Gate Previous Year Question Papers For Chemistry Free and Paid eBooks
  - Gate Previous Year Question Papers For Chemistry Public Domain eBooks
  - Gate Previous Year Question Papers For Chemistry eBook Subscription Services
  - Gate Previous Year Question Papers For Chemistry Budget-Friendly Options
6. Navigating Gate Previous Year Question Papers For Chemistry eBook Formats
  - ePub, PDF, MOBI, and More
  - Gate Previous Year Question Papers For Chemistry Compatibility with Devices
  - Gate Previous Year Question Papers For Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Gate Previous Year Question Papers For Chemistry
  - Highlighting and Note-Taking Gate Previous Year Question Papers For Chemistry
  - Interactive Elements Gate Previous Year Question Papers For Chemistry
8. Staying Engaged with Gate Previous Year Question Papers For Chemistry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gate Previous Year Question Papers For Chemistry
9. Balancing eBooks and Physical Books Gate Previous Year Question Papers For Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gate Previous Year Question Papers For Chemistry
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Gate Previous Year Question Papers For Chemistry
  - Setting Reading Goals Gate Previous Year Question Papers For Chemistry
  - Carving Out Dedicated Reading Time

- 
12. Sourcing Reliable Information of Gate Previous Year Question Papers For Chemistry
    - Fact-Checking eBook Content of Gate Previous Year Question Papers For Chemistry
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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