

⑤ 10 pts.

③ 20 pts.

① 40 pts.

① 50 pts.

Midterm Study Guide

50 min.

10 questions.

Jiahao Shen

11:15 - 12:05.

October 31, 2023

11:10 8/8%

10 questions in total. Material includes everything before the midterm.

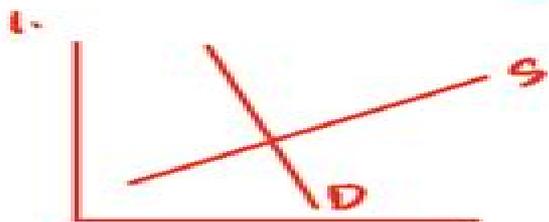
Module 1

542 / Module 3

Module 2

134.

1. How to compare elasticity of Demand and Supply in a graph? How does S/D shift affect equilibrium P and Q?
- ② Given a table with Q/TC/MR/... , how to find different costs for each quantity? How does firm decide profit maximization quantity? $P = MC$.
3. How does different return to scale imply pattern of total cost?
4. What is complement and substitute goods? How does price change of one good affect demand for the other good?
5. How does different elasticity affect change in total revenue as price changes?
6. How to find equilibrium in a S/D graph. How to determine size of excess demand / supply. How to find CS and PS in a graph.
7. Interpret the relationship between elasticity and total revenue. For a good, what can we say about elasticity of different brands for such good?
8. How does consumer maximize utility? What is the equation? $\frac{MU_1}{P_1} = \frac{MU_2}{P_2}$.
9. Interpret elasticity. What factors determine whether a good is elastic or inelastic? $\epsilon = -1.5$. ① what is elasticity ② percentage change.
10. How does shift of S and D affect equilibrium P and Q?



S is more elastic
比D 斜率低.

$|\epsilon| > 1$, elastic.
 $|\epsilon| < 1$, inelastic.

1.0 是正数, 平数点
increase / decrease

If the price elasticity is equal to -1.5, it means that the quantity of a product's demand has increased by 15% in response to a 10% increase

It Essentials Midterm Study Guide

Walter Crosby Eells



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Essentials for Blended Learning Jared Stein, Charles R. Graham, 2014-01-03 *Essentials for Blended Learning A Standards Based Guide* provides a practical streamlined approach for creating effective learning experiences by blending online activities and the best of face to face teaching This guide is Easy to use Clear jargon free writing illustrations and references to online resources help readers understand concepts Streamlined A simple but effective design process focuses on creating manageable activities for the right environment Practical Real world examples from different subject areas help teachers understand principles in context Contemporary The variety of modern connected technologies covered in the guide addresses a range of teaching challenges Forward Looking The approach bridges the gap between formal classroom learning and informal lifelong learning Standards based Guidelines and standards are based on current research in the field relevant learning theories and practitioner experiences Effective blended learning requires significant rethinking of teaching practices and a fundamental redesign of course structure *Essentials for Blended Learning A Standards Based Guide* simplifies these difficult challenges without neglecting important opportunities to transform teaching This guide is suitable for teachers in any content area Please visit www.essentialforblended.com for additional resources **Instructor's**

Manual for Essentials of Managerial Finance, Third Edition John Fred Weston, Eugene F. Brigham, 1974 [Essentials of Physical Chemistry](#) Don Shillady, 2011-07-27 At a time when U S high school students are producing low scores in mathematics and science on international examinations a thorough grounding in physical chemistry should not be considered optional for science undergraduates Based on the author s thirty years of teaching *Essentials of Physical Chemistry* merges coverage of calculus with chemistry and molecular physics in a friendly yet thorough manner Reflecting the latest ACS guidelines the book can be used as a one or two semester course and includes special topics suitable for senior projects The book begins with a math and physics review to ensure all students start on the same level and then discusses the basics of thermodynamics and kinetics with mathematics tuned to a level that stretches students abilities It then provides material for an optional second semester course that shows students how to apply their enhanced mathematical skills in a brief historical development of the quantum mechanics of molecules Emphasizing spectroscopy the text is built on a foundation of quantum chemistry and more mathematical detail and examples It contains sample classroom tested exams to gauge how well students know how to use relevant formulas and to display successful understanding of key concepts Coupling the development of mathematical skills with chemistry concepts encourages students to learn mathematical derivations Mini biographies of famous scientists make the presentation more interesting from a people point of view Stating the basic concepts of quantum chemistry in terms of analogies provides a pedagogically useful technique Covering key topics such as the critical point of a van der Waals gas the Michaelis Menten equation and the entropy of mixing this classroom tested text highlights applications across the range of chemistry forensic science pre medical science and chemical engineering In a

presentation of fundamental topics held together by clearly established mathematical models the book supplies a quantitative discussion of the merged science of physical chemistry

Student Study Guide to Accompany Essentials of Psychology Feldman,2002-04

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Essentials of Planning and Evaluation for Public Health Karen M. Perrin,Perrin,2014-09-17 Case Study Healthy Food Healthy Students HFHS Goal Statement Objectives Case Study Discussion Questions Student Activities References Chapter 2 Ethics Chapter Objectives Key Terms Introduction Historical Background of Ethical Principles Basic Ethical Principles Ethical Links between Research and Evaluation Institutional Review Board Informed Consent Risk Benefit Assessment Selection of Individuals and Special Populations Ethical Guidelines for Evaluators Challenges Faced by Evaluators Confidentiality of Personal Information Summary

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