

# Mass Relationships in Chemical Reactions

Group 2 STEM 12 - Euclid

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## INTRODUCTION

A reaction is said to have been completed if one of the reactants is completely consumed by the reaction. In this experiment, sodium bicarbonate (baking soda) is made to react with hydrochloric acid to produce sodium chloride according to the reaction:



You will use an accurately measured amount of  $\text{NaHCO}_3$  and add enough  $\text{HCl}$  to totally drain the bicarbonate. You will separate  $\text{NaCl}$  from the other products and determine its mass. This is the reaction's actual yield. Using the mass relationships in the balanced chemical equation above, the theoretical yield may be computed. The percentage yield can be calculated by dividing the actual yield by the theoretical yield.

## MATERIALS

- Evaporating dish
- Watch glass



# Mass Relationship In Chemical Reactions

**Jicheng Xie**



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