

Stoichiometry: Mole to Mass Problems

1.  $\text{Mg}(s) + 2\text{HCl}(aq) \rightarrow \text{MgCl}_2(aq) + \text{H}_2(g)$
- How many grams of HCl are consumed by the reaction of 2.50 moles of magnesium?
  - What is the mass in grams of  $\text{H}_2$  gas when 4.0 moles of HCl is added to the reaction?

2. Acetylene gas ( $\text{C}_2\text{H}_2$ ) is produced as a result of the following reaction.



- If 3.20 moles of  $\text{CaC}_2$  are consumed in this reaction, how many grams of  $\text{H}_2\text{O}$  are needed?
  - How many grams of  $\text{Ca}(\text{OH})_2$  would be formed with 3.20 moles of  $\text{CaC}_2$ ?
3. Acetylene gas,  $\text{C}_2\text{H}_2$ , is used in welding, produces an extremely hot flame when it burns in pure oxygen according to the following reaction.



How many moles of water ( $\text{H}_2\text{O}$ ) are produced when 25.0 grams of  $\text{C}_2\text{H}_2$  burns completely?

4.  $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$

How many moles of iron, Fe, are produced with 25.0 grams of magnesium, Mg?

5. Laughing gas (nitrous oxide,  $\text{N}_2\text{O}$ ) is sometimes used as an anesthetic in dentistry.



- How many moles of  $\text{NH}_4\text{NO}_3$  are required to produce 33.0g of  $\text{N}_2\text{O}$ ?
- How many moles of water are produced with 45.0g of  $\text{N}_2\text{O}$ ?

Answers to Stoichiometry: Mole to Mass Problems

# Mass To Mass Stoichiometry Problems Chemfiesta

**David Callaghan**



## **Mass To Mass Stoichiometry Problems Chemfiesta:**

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