

Density Worksheet

Physical Science

$D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

1. The ratio of an object's mass to its _____ is called the *density* of the object.

2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.

3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - a. mass = 453 g; volume = 225 cm³
 - b. mass = 5.0 g; volume = 10.0 cm³
 - c. mass = 26.1 g; volume = 2.0 mL

4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.

5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Gravity And Acceleration Physical Science If8767 Answers

S Ben Porath



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Table of Contents Gravity And Acceleration Physical Science If8767 Answers

1. Understanding the eBook Gravity And Acceleration Physical Science If8767 Answers
 - The Rise of Digital Reading Gravity And Acceleration Physical Science If8767 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gravity And Acceleration Physical Science If8767 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gravity And Acceleration Physical Science If8767 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gravity And Acceleration Physical Science If8767 Answers
 - Personalized Recommendations
 - Gravity And Acceleration Physical Science If8767 Answers User Reviews and Ratings
 - Gravity And Acceleration Physical Science If8767 Answers and Bestseller Lists
5. Accessing Gravity And Acceleration Physical Science If8767 Answers Free and Paid eBooks
 - Gravity And Acceleration Physical Science If8767 Answers Public Domain eBooks
 - Gravity And Acceleration Physical Science If8767 Answers eBook Subscription Services

- Gravity And Acceleration Physical Science If8767 Answers Budget-Friendly Options
- 6. Navigating Gravity And Acceleration Physical Science If8767 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gravity And Acceleration Physical Science If8767 Answers Compatibility with Devices
 - Gravity And Acceleration Physical Science If8767 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gravity And Acceleration Physical Science If8767 Answers
 - Highlighting and Note-Taking Gravity And Acceleration Physical Science If8767 Answers
 - Interactive Elements Gravity And Acceleration Physical Science If8767 Answers
- 8. Staying Engaged with Gravity And Acceleration Physical Science If8767 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gravity And Acceleration Physical Science If8767 Answers
- 9. Balancing eBooks and Physical Books Gravity And Acceleration Physical Science If8767 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gravity And Acceleration Physical Science If8767 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gravity And Acceleration Physical Science If8767 Answers
 - Setting Reading Goals Gravity And Acceleration Physical Science If8767 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gravity And Acceleration Physical Science If8767 Answers
 - Fact-Checking eBook Content of Gravity And Acceleration Physical Science If8767 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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