

# The Periodic Table of the Elements

|                    |                    |                   |                    |                    |                    |                    |                    |                    |                    |                    |                    |                    |                     |                     |                     |                     |                     |                     |    |    |
|--------------------|--------------------|-------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|---------------------|---------------------|---------------------|---------------------|---------------------|---------------------|----|----|
| 1                  |                    |                   |                    |                    |                    |                    |                    |                    |                    |                    |                    | 18                 |                     |                     |                     |                     |                     |                     |    |    |
| 1<br>H<br>1.01     |                    |                   |                    |                    |                    |                    |                    |                    |                    |                    |                    | 1<br>He<br>4.00    |                     |                     |                     |                     |                     |                     |    |    |
| 2                  | 3                  | 4                 |                    |                    |                    |                    |                    |                    |                    |                    |                    |                    | 5                   | 6                   | 7                   | 8                   | 9                   | 10                  | 11 | 12 |
| 3<br>Li<br>6.94    | 4<br>Be<br>9.01    |                   |                    |                    |                    |                    |                    |                    |                    |                    |                    | 5<br>B<br>10.81    | 6<br>C<br>12.01     | 7<br>N<br>14.01     | 8<br>O<br>16.00     | 9<br>F<br>19.00     | 10<br>Ne<br>20.18   |                     |    |    |
| 11<br>Na<br>22.99  | 12<br>Mg<br>24.31  |                   |                    |                    |                    |                    |                    |                    |                    |                    |                    | 13<br>Al<br>26.98  | 14<br>Si<br>28.09   | 15<br>P<br>30.97    | 16<br>S<br>32.07    | 17<br>Cl<br>35.45   | 18<br>Ar<br>39.95   |                     |    |    |
| 19<br>K<br>39.10   | 20<br>Ca<br>40.08  | 21<br>Sc<br>44.96 | 22<br>Ti<br>47.88  | 23<br>V<br>50.94   | 24<br>Cr<br>52.00  | 25<br>Mn<br>54.94  | 26<br>Fe<br>55.85  | 27<br>Co<br>58.93  | 28<br>Ni<br>58.69  | 29<br>Cu<br>63.55  | 30<br>Zn<br>65.39  | 31<br>Ga<br>69.72  | 32<br>Ge<br>72.61   | 33<br>As<br>74.92   | 34<br>Se<br>78.96   | 35<br>Br<br>79.90   | 36<br>Kr<br>83.80   |                     |    |    |
| 37<br>Rb<br>85.47  | 38<br>Sr<br>87.62  | 39<br>Y<br>88.91  | 40<br>Zr<br>91.22  | 41<br>Nb<br>92.91  | 42<br>Mo<br>95.94  | 43<br>Tc<br>(98)   | 44<br>Ru<br>101.07 | 45<br>Rh<br>102.91 | 46<br>Pd<br>106.42 | 47<br>Ag<br>107.87 | 48<br>Cd<br>112.41 | 49<br>In<br>114.82 | 50<br>Sn<br>118.71  | 51<br>Sb<br>121.76  | 52<br>Te<br>127.60  | 53<br>I<br>126.90   | 54<br>Xe<br>131.29  |                     |    |    |
| 55<br>Cs<br>132.91 | 56<br>Ba<br>137.33 | 57-70<br>*        | 71<br>Lu<br>174.97 | 72<br>Hf<br>178.49 | 73<br>Ta<br>180.95 | 74<br>W<br>183.84  | 75<br>Re<br>186.21 | 76<br>Os<br>190.23 | 77<br>Ir<br>192.22 | 78<br>Pt<br>195.08 | 79<br>Au<br>196.97 | 80<br>Hg<br>200.59 | 81<br>Tl<br>204.38  | 82<br>Pb<br>207.20  | 83<br>Bi<br>208.98  | 84<br>Po<br>(209)   | 85<br>At<br>(210)   | 86<br>Rn<br>(222)   |    |    |
| 87<br>Fr<br>(223)  | 88<br>Ra<br>(226)  | 89-102<br>**      | 103<br>Lr<br>(262) | 104<br>Rf<br>(261) | 105<br>Db<br>(268) | 106<br>Sg<br>(271) | 107<br>Bh<br>(272) | 108<br>Hs<br>(278) | 109<br>Mt<br>(276) | 110<br>Ds<br>(285) | 111<br>Rg<br>(284) | 112<br>Cn<br>(285) | 113<br>Uut<br>(284) | 114<br>Uuq<br>(289) | 115<br>Uup<br>(288) | 116<br>Uuh<br>(289) | 117<br>Uus<br>(289) | 118<br>Uuo<br>(294) |    |    |

- Alkali metals
- Alkaline earth metals
- Transition metals
- Other metals
- Metalloids (semi-metal)
- Nonmetals
- Halogens
- Noble gases
- Lanthanides
- Actinides

Element name → Mercury

Atomic # ← 80

Symbol → Hg

Avg. Mass ← 200.59

|              |                    |                    |                    |                    |                   |                    |                    |                    |                    |                    |                    |                    |                    |                    |
|--------------|--------------------|--------------------|--------------------|--------------------|-------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|
| *lanthanides | 57<br>La<br>138.91 | 58<br>Ce<br>140.12 | 59<br>Pr<br>140.91 | 60<br>Nd<br>144.24 | 61<br>Pm<br>(145) | 62<br>Sm<br>150.36 | 63<br>Eu<br>151.97 | 64<br>Gd<br>157.25 | 65<br>Tb<br>158.93 | 66<br>Dy<br>162.50 | 67<br>Ho<br>164.93 | 68<br>Er<br>167.26 | 69<br>Tm<br>168.93 | 70<br>Yb<br>173.04 |
| **actinides  | 89<br>Ac<br>(227)  | 90<br>Th<br>232.04 | 91<br>Pa<br>231.04 | 92<br>U<br>238.03  | 93<br>Np<br>(237) | 94<br>Pu<br>(244)  | 95<br>Am<br>(243)  | 96<br>Cm<br>(247)  | 97<br>Bk<br>(247)  | 98<br>Cf<br>(251)  | 99<br>Es<br>(252)  | 100<br>Fm<br>(257) | 101<br>Md<br>(258) | 102<br>No<br>(259) |

# Groups Of The Periodic Table Answer Key

**Michael Brown**



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