

**Naming Ionic Compounds Practice Worksheet - Solutions**

*Name the following ionic compounds:*

- |     |                              |                                |
|-----|------------------------------|--------------------------------|
| 1)  | $\text{NH}_4\text{Cl}$       | <b>ammonium chloride</b>       |
| 2)  | $\text{Fe}(\text{NO}_3)_3$   | <b>iron (III) nitrate</b>      |
| 3)  | $\text{TiBr}_3$              | <b>titanium (III) bromide</b>  |
| 4)  | $\text{Cu}_3\text{P}$        | <b>copper (I) phosphide</b>    |
| 5)  | $\text{SnSe}_2$              | <b>tin (IV) selenide</b>       |
| 6)  | $\text{GaAs}$                | <b>gallium arsenide</b>        |
| 7)  | $\text{Pb}(\text{SO}_4)_2$   | <b>lead (IV) sulfate</b>       |
| 8)  | $\text{Be}(\text{HCO}_3)_2$  | <b>beryllium bicarbonate</b>   |
| 9)  | $\text{Mn}_2(\text{SO}_3)_3$ | <b>manganese (III) sulfite</b> |
| 10) | $\text{Al}(\text{CN})_3$     | <b>aluminum cyanide</b>        |

*Write the formulas for the following compounds:*

- |     |                          |   |
|-----|--------------------------|---|
| 11) | chromium (VI) phosphate  | <b><math>\text{Cr}(\text{PO}_4)_2</math></b>                    |
| 12) | vanadium (IV) carbonate  | <b><math>\text{V}(\text{CO}_3)_2</math></b>                     |
| 13) | tin (II) nitrite         | <b><math>\text{Sn}(\text{NO}_2)_2</math></b>                    |
| 14) | cobalt (III) oxide       | <b><math>\text{Co}_2\text{O}_3</math></b>                       |
| 15) | titanium (II) acetate    | <b><math>\text{Ti}(\text{C}_2\text{H}_3\text{O}_2)_2</math></b> |
| 16) | vanadium (V) sulfide     | <b><math>\text{V}_2\text{S}_5</math></b>                        |
| 17) | chromium (III) hydroxide | <b><math>\text{Cr}(\text{OH})_3</math></b>                      |
| 18) | lithium iodide           | <b><math>\text{LiI}</math></b>                                  |
| 19) | lead (II) nitride        | <b><math>\text{Pb}_3\text{N}_2</math></b>                       |
| 20) | silver bromide           | <b><math>\text{AgBr}</math></b>                                 |

# Naming Ionic Compounds Practice Problems Answer Key

**Leo J. Malone, Theodore O. Dolter**



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