

## MULTIPLE CHOICE QUESTIONS

1. Reduction always takes place: (LHR 05, 12)  
 (a) At anode (b) At cathode  
 (c) At both electrodes (d) does not occur at electrodes
2. The electrolyte KOH is used in cell: (GRW 06)  
 (a) Lead accumulator (b) Ni – Cd cell  
 (c) Alkaline battery (d) Silver oxide battery
3. In the reaction  $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$ : (GRW 10, GRW 09, LHR 06)  
 (a) Fe is reduced (b) Fe is oxidized  
 (c)  $\text{Cl}_2$  is oxidized (d) None of these happens
4. Oxidation state of hydrogen in  $\text{CaH}_2$  is: (GRW 07)  
 (a) +1 (b) -1 (c) +2 (d) zero
5. Cu metal can be purified in electrolytic cell by making the impure Cu as: (GRW 07)  
 (a) Anode (b) Cathode (c) Anode and Cathode (d) Depends upon nature of solution
6. Loss of electrons is called: (LHR 07)  
 (a) Oxidation (b) Reduction (c) Hydration (d) Dehydration
7. Fuel cells convert chemical energy into: (GRW 07)  
 (a) Heat energy (b) Light energy (c) Electrical energy (d) Mechanical energy
8. Electrolysis is used for: (LHR 08)  
 (a) Electroplating (b) Manufacture of sodium metal (c) Manufacture of Al (d) All of these
9. Nelson's cell and Down's cell are example of: (PSD 09, GRW 08)  
 (a) Electrochemical cell (b) Galvanic cell (c) Electrolytic cell (d) None of these
10. The oxidation number of Cr in  $\text{K}_2\text{Cr}_2\text{O}_7$  is: (GRW 09, 10, LHR 10, 12, 13)  
 (a) +14 (b) +12 (c) +6 (d) +13
11. The electrolyte used in fuel cell is: (LHR 09, DGG, GRW 10)  
 (a) Aqueous  $\text{NaCl}$  (b) Molten  $\text{NaCl}$  (c)  $\text{KOH}$  (d)  $\text{H}_2\text{SO}_4$

# Electrochemistry Multiple Multiple Choice Questions

**Jianjun Gao**



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