

Unit 5 Reactivity

Topic 5.1 Reactivity and displacement reactions

Exercise 5.1A Using the reactivity series

- 1 less vigorously
- 2 It does not react at all; give credit for silver reacts less / is less reactive than copper.
- 3 Zinc will react more quickly than iron with dilute acid.
- 4 No; there will be not be a displacement reaction because copper is less reactive than iron.
- 5 Silver or gold; give credit if another unreactive metal such as platinum is named.
- 6 iron, lead or copper
- 7
 - a yes
 - b no
 - c no
 - d yes
 - e yes
- 8 $\text{magnesium} + \text{zinc chloride} \rightarrow \text{magnesium} + \text{zinc chloride}$

Exercise 5.1B Displacement reactions

- 1
 - a It tells you that zinc is more reactive than copper.
 - b $\text{zinc} + \text{copper sulfate} \rightarrow \text{zinc sulfate} + \text{copper}$
 - c It tells you that zinc is less reactive than magnesium.
 - d magnesium (most reactive), iron, copper (least reactive).
- 2
 - a Sofia could place a small piece of the metal in each of the solutions. She will be able to observe the reactions with the solutions and if there is a reaction with

copper sulfate she will know that the metal is more reactive than copper. If there is no reaction with zinc sulfate she will know that it is less reactive than zinc. By observing the reactions in all four of the solutions she will be able to work out which metal she has.

- b To distinguish between iron and zinc you could place a piece of each of the two metals in a solution of iron sulfate. The zinc would react but the iron would not. Credit the use of any iron salt.

Exercise 5.1C Displacing metals

- 1 Metal E is the most reactive because it has the most displacement reactions.

2

		Metal					
		A	B	C	D	E	F
Metal salt	A	X	✓	X	✓	✓	✓
	B	X	X	X	X	✓	X
	C	✓	✓	X	✓	✓	✓
	D	X	✓	X	X	✓	X
	E	X	X	X	X	X	X
	F	X	✓	X	✓	✓	X

- 3 The metals that are more reactive than C, copper, and less reactive than D, zinc, are lead and iron. Iron is more reactive than lead, so **A is lead** and **F is iron**. The metals that are more reactive than D, zinc, are magnesium and aluminium. B is less reactive than E, so **B is aluminium** and **E is magnesium**. Learners should not suggest an alkali metal as these are

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