

QUESTION – CHEMISTRY

FORM -2

CHEMICAL FAMILIES

1. Study the information in the table below and answer the questions that follow:

Element	Atomic radius (nm)	Ionic radius (nm)
W	0.114	0.195
X	0.072	0.136
Y	0.133	0.216
Z	0.099	0.181

(a) Would these form part of a metallic or a non-metallic group? Explain

(b) Suggest an element in the table above likely to be the most reactive. Explain

2. State the reason for using Argon in electric light bulbs

3. Study the information in the table below and answer the questions that follow. The letters do not represent the actual symbols of the elements.

Element	Electronic configuration	Boiling point
X	2.7	-188°C
Y	2.8.7	-35°C
Z	2.8.8.7	59°C

(a) What is the general name given to the group in which the elements X, Y and Z belong?

Form Chemistry Questions

M Walker



Form Chemistry Questions:

This book delves into Form Chemistry Questions. Form Chemistry Questions is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Form Chemistry Questions, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Form Chemistry Questions
 - Chapter 2: Essential Elements of Form Chemistry Questions
 - Chapter 3: Form Chemistry Questions in Everyday Life
 - Chapter 4: Form Chemistry Questions in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, this book will provide an overview of Form Chemistry Questions. This chapter will explore what Form Chemistry Questions is, why Form Chemistry Questions is vital, and how to effectively learn about Form Chemistry Questions.
3. In chapter 2, this book will delve into the foundational concepts of Form Chemistry Questions. The second chapter will elucidate the essential principles that need to be understood to grasp Form Chemistry Questions in its entirety.
4. In chapter 3, the author will examine the practical applications of Form Chemistry Questions in daily life. This chapter will showcase real-world examples of how Form Chemistry Questions can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Form Chemistry Questions in specific contexts. This chapter will explore how Form Chemistry Questions is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Form Chemistry Questions. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Form Chemistry Questions.

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