

# Polyatomic Ions Worksheet

## Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	_____	$\text{CO}_3^{2-}$	_____
nitrite	_____	$\text{MnO}_3^{1-}$	_____
perphosphate	_____	$\text{SO}_5^{2-}$	_____
hypoiodite	_____	$\text{BrO}_2^{1-}$	_____
chlorite	_____	$\text{CO}_4^{2-}$	_____
phosphite	_____	$\text{PO}_5^{3-}$	_____
percarbonate	_____	$\text{ClO}^{1-}$	_____
bromate	_____	$\text{IO}_2^{1-}$	_____
hyposulfite	_____	$\text{PO}_4^{3-}$	_____
permanganate	_____	$\text{NO}_2^{1-}$	_____
carbonite	_____	$\text{SO}_4^{2-}$	_____

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	_____	$\text{Ba}(\text{IO}_3)_2$	_____
sodium nitrite	_____	$\text{Ga}(\text{CNO})_3$	_____
ammonium chloride	_____	$\text{Ag}_2\text{SO}_3$	_____
calcium bisulfate	_____	$\text{MgCO}$	_____
rubidium perchlorate	_____	$\text{NH}_3\text{NO}_2$	_____
strontium sulfite	_____	$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$	_____
aluminum acetate	_____	$\text{SrSO}_5$	_____
ammonium nitrate	_____	$\text{RbClO}_2$	_____
magnesium hypocarbonite	_____	$\text{Ca}(\text{HSO}_4)_2$	_____
silver hyposulfite	_____	$\text{NH}_4\text{Cl}$	_____
gallium cyanate	_____	$\text{NaNO}$	_____
barium iodate	_____	$\text{Be}(\text{OH})_2$	_____

# Formulas With Polyatomic Ions Worksheet Answers

## Page 57

**K Payea**



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