

LIMITING REACTANT STOICHIOMETRY WORKSHEET

Name _____	Section _____	Date _____
Limiting Reactants		
Directions: Determine the limiting reactant in each problem.		
1) In $2\text{NaOH} + \text{MgCl}_2 \rightarrow \text{Mg(OH)}_2 + 2\text{NaCl}$, 5.818 grams of NaOH react with 2.712 grams of MgCl_2 .		
2) 0.0718 grams of HCl react with 0.0002 grams of Ca(OH)_2 in the reaction $2\text{HCl} + \text{Ca(OH)}_2 \rightarrow 2\text{H}_2\text{O} + \text{CaCl}_2$.		
3) 0.2000 grams of Nitrogen in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ combine with 0.2000 grams of hydrogen.		

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1) In $2\text{NaOH} + \text{MgCl}_2 \rightarrow \text{Mg(OH)}_2 + 2\text{NaCl}$, 5.818 grams of NaOH react with 2.712 grams of MgCl_2 .		
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3) 0.2000 grams of Nitrogen in the reaction $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ combine with 0.2000 grams of hydrogen.		
4) In the reaction $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$, 10.000 grams of O_2 react with 5.968 grams of C_2H_2 .		
5) In the reaction $2\text{Al(OH)}_3 + 3\text{H}_2\text{CO}_3 \rightarrow \text{Al}_2(\text{CO}_3)_3 + 6\text{H}_2\text{O}$, 1.18 grams of Al(OH)_3 and 1.312 grams of H_2CO_3 combine.		
6) 0.838 grams of sodium and 0.7021 grams of sodium nitrate react in the following process: $10\text{Na} + 2\text{NaNO}_3 \rightarrow 6\text{Na}_2\text{O} + \text{N}_2$.		
7) 0.0700 grams of AsH_3 react with 1.201 grams of H_2SO_4 to produce products in the following reaction: $2\text{AsH}_3 + 3\text{H}_2\text{SO}_4 \rightarrow 2\text{AsH} + \text{Al}_2(\text{PO}_4)_3$.		
8) When K_3PO_4 and NaOH react in $\text{K}_3\text{PO}_4 + 3\text{NaOH} \rightarrow 3\text{KOH} + \text{Na}_3\text{PO}_4$, 5.41 grams of NaOH react with 3.88 grams K_3PO_4 .		
9) In $\text{Pb(NO}_3)_2 + \text{H}_2\text{S} \rightarrow \text{PbS} + \text{HNO}_3$, 10.02 grams of $\text{Pb(NO}_3)_2$ react with 7.88 grams of H_2S .		
10) 0.4000 grams of H_2SO_4 combine with 0.2880 grams of SO_2 in the reaction $2\text{H}_2\text{SO}_4 + 2\text{SO}_2 \rightarrow 2\text{S}_2\text{O}_8^{2-} + 2\text{H}_2\text{O}$.		

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A CHEMISTRY WORKSHEET

10 PROBLEMS

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PT Brinkman



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