

Student Exploration: Limiting Reactants

Essential Question: How do you determine the limiting reactant in a chemical reaction, and how does it affect the amount of product that is formed?

What You Will Do: In this exploration, you will use the Gizmo to explore how the limiting reactant affects the amount of product that is formed. You will also explore how the limiting reactant affects the amount of product that is formed.

1. How many molecules of H_2 are there? _____ How many molecules of O_2 are there? _____

2. How many molecules of H_2O are there? _____ How many molecules of H_2 are there? _____

3. How many molecules of H_2O are there? _____ How many molecules of H_2 are there? _____

What is happening?

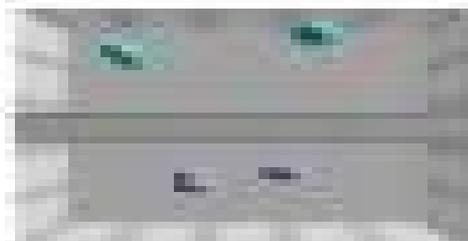
Just as ingredients can be used together to make a limited substance (such as a cake), a limited amount of reactants can be used to make a limited amount of product. The reactants that control the amount of product are called **limiting reactants**. The other reactants are called **excess reactants**.

Reactants that are **limiting reactants** are used up first. Reactants that are **excess reactants** are used up last. Reactants that are **limiting reactants** are used up first. Reactants that are **excess reactants** are used up last.

The limiting reactant is the reactant that is used up first. The amount of product that is formed is limited by the amount of limiting reactant that is used up.

4. How many molecules of H_2 are there? _____ How many molecules of O_2 are there? _____

5. How many molecules of H_2O are there? _____ How many molecules of H_2 are there? _____



Limiting Reactant Student Exploration Sheet Answer

G Orfield



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Proceedings American Society for Engineering Education. Conference,1994 **Annual Conference Proceedings**
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security Founded by Manhattan Project Scientists the Bulletin s iconic Doomsday Clock stimulates solutions for a safer world
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Decoding **Limiting Reactant Student Exploration Sheet Answer**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Limiting Reactant Student Exploration Sheet Answer**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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